

## Reaction Rates & Enthalpy Test SCH4U #13 Name:

K T C A  
(Product) 14 12 12 14

### Assessment of Learning

Multiple Choice. Circle the most correct answer. (14K, 6T) 1-14 are K and 15-20 are T

1. For an endothermic reaction, more energy is:

- A) released in forming bonds than is absorbed in breaking bonds
- B) released in breaking bonds than is absorbed in forming bonds
- C) absorbed in forming bonds than is released in breaking bonds
- D) absorbed in breaking bonds than is released in forming bonds
- E) none of the above

2. Of the following, \_\_\_\_\_ is an exothermic process.

- A) melting
- B) subliming
- C) freezing
- D) boiling
- E) All of the above are exothermic

3. Which statements are true?

- i) A thermochemical equation must include the physical states of all substances.
- ii) The coefficients in a thermochemical equation are interpreted as moles.
- iii) The sign of  $\Delta H$  for an endothermic reaction is negative.
- A) i) and ii)
- B) i) and iii)
- C) ii) and iii)
- D) i), ii), and iii)

4. As the number of covalent bonds between two atoms increases, the distance between the atoms \_\_\_\_\_ and the strength of the bond between them

- A) increases, increases
- B) decreases, decreases
- C) increases, decreases
- D) decreases, increases
- E) is unpredictable

5. For which of the following substances does  $\Delta H_f = 0$ ?

- A)  $\text{CO}_2(\text{g})$
- B)  $\text{H}_2\text{O}(\text{g})$
- C)  $\text{Na}(\text{s})$
- D)  $\text{Br}_2(\text{g})$
- E)  $\text{C}(\text{diamond})$

6. If two substances of equal mass are supplied the same amount of thermal energy and their final temperatures are different, you can assume that they have

- A) different heat capacities
- B) the same volume
- C) different abilities to conduct heat
- D) different densities
- E) different volumes

7. Of the following, \_\_\_\_\_ will lower the activation energy for a reaction.

- A) increasing the concentrations of reactants
- B) raising the temperature of the reaction

# Answers For Sch4u Unit 4

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