

Reaction Rates & Enthalpy Test SCH4U #13 Name:

____K ____T ____C ____A
(Product)
14 12 12 14

Assessment of Learning

Multiple Choice: Circle the most correct answer. (14K, 6T) 1-14 are K and 15-20 are T

1. For an endothermic reaction, more energy is:

- A) released in forming bonds than is absorbed in breaking bonds
- B) released in breaking bonds than is absorbed in forming bonds
- C) absorbed in forming bonds than is released in breaking bonds
- D) absorbed in breaking bonds than is released in forming bonds
- E) none of the above

2. Of the following, _____ is an exothermic process.

- A) melting B) subliming C) freezing D) boiling E) All of the above are exothermic

3. Which statements are true?

- i) A thermochemical equation must include the physical states of all substances.
 - ii) The coefficients in a thermochemical equation are interpreted as moles.
 - iii) The sign of ΔH for an endothermic reaction is negative.
- A) i) and ii) B) i) and iii) C) ii) and iii) D) i), ii), and iii)

4. As the number of covalent bonds between two atoms increases, the distance between the atoms _____ and the strength of the bond between them _____.

- A) increases, increases B) decreases, decreases C) increases, decreases
- D) decreases, increases E) is unpredictable

5. For which of the following substances does $\Delta H_f = 0$?

- A) $\text{CO}_2(\text{g})$ B) $\text{H}_2\text{O}(\text{g})$ C) $\text{Na}(\text{s})$ D) $\text{Br}_2(\text{g})$ E) C(diamond)

6. If two substances of equal mass are supplied the same amount of thermal energy and their final temperatures are different, you can assume that they have

- A) different heat capacities B) the same volume C) different abilities to conduct heat
- D) different densities E) different volumes

7. Of the following, _____ will lower the activation energy for a reaction.

- A) increasing the concentrations of reactants B) raising the temperature of the reaction

Answers For Sch4u Unit 4

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