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## **Medical Laboratory Science Review - Harr – Chemistry |81 questions with correct answers**

Which formula correctly describes the relationship between absorbance and %T ?

A.  $A = 2 - \log \%T$

B.  $A = \log 1/T$

C.  $A = -\log T$

D. All of these options Answer ✓✓ D Absorbance is proportional to the inverse log of transmittance.

$$A = -\log T = \log 1/T$$

Multiplying the numerator and denominator by 100 gives:

$$A = \log (100/100 \times T)$$

$$100 \times T = \%T, \text{ substituting } \%T$$

for  $100 \times T$  gives:

$$A = \log 100/\%T$$

$$A = \log 100 - \log \%T$$

$$A = 2.0 - \log \%T$$

For example, if  $\%T = 10.0$ , then:

$$A = 2.0 - \log 10.0$$

$$\log 10.0 = 1.0$$

$$A = 2.0 - 1.0 = 1.0$$

A solution that has a transmittance of 1.0 %T

# Answers For Lab 6b Heath Chemistry

**Michael Seilmaier**

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