

Atoms PhET Lab

Introduction: Atoms are the smallest things that retain the properties of matter we can observe. Atoms are made of three subatomic particles: protons, neutrons, and electrons.

- Protons have a mass of _____ amu and a charge of _____.
- Neutrons have a mass of _____ amu and a charge of _____.
- Electrons have a mass of nearly _____ amu and a charge of _____.

In this simulation, you will build atoms, subatomic particle by subatomic particle and observe the effect of adding more of each particle. When the subatomic particles in an atom change, an ion, isotope or different element will be created.

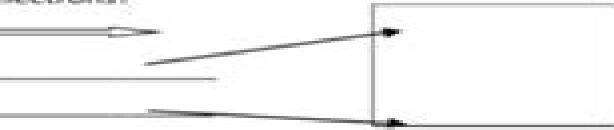
**Procedure 1: Play with the Sims → Chemistry → Build An Atom** [Run Now!](#)

Begin by playing with the simulation for a while. Become familiar with the interface. What happens when you add protons, neutrons, or electrons? To start over, click **Reset All**.



Show the symbol, atomic mass, and charge by clicking on the .

Analysis Questions

- Ions are atoms of the same element with different numbers of _____.
- Isotopes are atoms of the same element with different numbers of _____.
- Adding or removing protons from an atom does what to the atom? _____.
- An atom with the same number of protons and electrons has a charge of _____.
- Adding two electrons to a neutral atom produces an ion with a charge of _____.
- An atom with six protons and five electrons would have a charge of _____.
- What atom is created with nine protons, nine neutrons, and nine electrons?
- Show the full symbol for the above atom in the box at the right. 
- What does the upper-left number in the symbol represent? _____
- What does the lower-left number in the symbol represent? _____

Game The Game

Click the tab at the top that says "Game".

Play each level until you make a perfect score. Call me over when you complete a level with a perfect score, and I will check it off.

Level 1 _____

Level 2 _____

Level 3 _____

Level 4 _____

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Overcoming Students' Misconceptions in Science Mageswary Karpudewan,Ahmad Nurulazam Md Zain,A.L. Chandrasegaran,2017-02-28 This book discusses the importance of identifying and addressing misconceptions for the successful teaching and learning of science across all levels of science education from elementary school to high school It suggests teaching approaches based on research data to address students common misconceptions Detailed descriptions of how these instructional approaches can be incorporated into teaching and learning science are also included The science education literature extensively documents the findings of studies about students misconceptions or alternative conceptions about various science concepts Furthermore some of the studies involve systematic approaches to not only creating but also implementing instructional programs to reduce the incidence of these misconceptions among high school science students These studies however are largely unavailable to classroom practitioners partly because they are usually found in various science education journals that teachers have no time to refer to or are not readily available to them

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