

**Assignment #3 – Compound Names and Formulas**  
**Single and Multi-valent ions**

A. Write the correct formula for the following compounds.

- |  |   |
|--|---|
| 1. gallium oxide <u><math>Ga_2O_3</math></u>       | 11. silicon oxide <u><math>SiO_2</math></u>       |
| 2. mercury (I) chloride <u><math>HgCl</math></u>   | 12. aluminum bromide <u><math>AlBr_3</math></u>   |
| 3. tin (IV) bromide <u><math>SnBr_4</math></u>     | 13. tin (II) sulphide <u><math>SnS</math></u>     |
| 4. indium oxide <u><math>In_2O_3</math></u>        | 14. germanium bromide <u><math>GeBr_4</math></u>  |
| 5. cadmium sulphide <u><math>CdS</math></u>        | 15. nickel (II) oxide <u><math>NiO</math></u>     |
| 6. magnesium oxide <u><math>MgO</math></u>         | 16. strontium astatide <u><math>SrAt_2</math></u> |
| 7. potassium nitride <u><math>K_3N</math></u>      | 17. aluminum oxide <u><math>Al_2O_3</math></u>    |
| 8. iron (III) sulphide <u><math>Fe_2S_3</math></u> | 18. copper (II) oxide <u><math>CuO</math></u>     |
| 9. gold (III) bromide <u><math>AuBr_3</math></u>   | 19. lawrencium fluoride <u><math>LrF_3</math></u> |
| 10. zinc oxide <u><math>ZnO</math></u>             | 20. actinium oxide <u><math>Ac_2O_3</math></u>    |

B. Name the following compounds correctly. Only include the Roman Numeral in the name if the first element has 2 or more combining capacities.

- |  |   |
|--|---|
| 1. $CaBr_2$ <u>Calcium Bromide</u>     | 9. $AsCl_5$ <u>Arsenic (V) Chloride</u>   |
| 2. $K_2O$ <u>Potassium Oxide</u>       | 10. $Eu_2O_3$ <u>Europium (III) Oxide</u> |
| 3. $Fe_2O_3$ <u>Iron (III) Oxide</u>   | 11. $NdBr_3$ <u>Neodymium Bromide</u>     |
| 4. $ZnF_2$ <u>Zinc Fluoride</u>        | 12. $Cu_3N_2$ <u>Copper (II) Nitride</u>  |
| 5. $CuCl$ <u>Copper (I) Chloride</u>   | 13. $CoS$ <u>Cobalt (II) Sulphide</u>     |
| 6. $Ni_2O_3$ <u>Nickel (III) Oxide</u> | 14. $ZrO_2$ <u>Zirconium Oxide</u>        |
| 7. $AuCl$ <u>Gold (I) Chloride</u>     | 15. $Li_2O$ <u>Lithium Oxide</u>          |
| 8. $HgBr$ <u>Mercury (I) Bromide</u>   | 16. $NaH$ <u>Sodium Hydride</u>           |

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**How Chemical Bonds Form and Chemical Reactions Proceed** V. I. Gankin, Yuri V. Gankin,1998

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