

Name: _____

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Chemistry 11

Classifying Matter

Assignment

1. Complete the following sentences by filling in the appropriate word from the list below.

Elements
Mixtures
Compound
Symbols
Properties

Carbon
Pure
Letters
Carbon dioxide

- a. _____ **Elements** _____ cannot be separated by physical or chemical means.
- b. Elements are abbreviated with _____ **symbols** _____, which consist of one or two _____ **letters** _____.
- c. Two or more elements chemically combined make up a _____ **compound** _____.
- d. _____ **Carbon** _____ is an example of an element and _____ **carbon dioxide** _____ is an example of a compound.
- e. Elements and compounds are called _____ **pure** _____ substances because they have a unique set of chemical and physical _____ **properties** _____.

2. Fill in the blanks in the following chart.

Element Name	Element Symbol	Common Ions Formed
a. Sodium	Na	+1
b. Copper	Cu	+2, +1
c. Tungsten	W	+6
d. Iron	Fe	+3, +2
e. Tin	Sn	+4, +2
f. Potassium	K	+1
g. Gold	Au	+3, +1
h. Silver	Ag	+1

3. Classify each of the following as one of an Atom (A), a Molecule (M) or an Ion (I).

a. P^{3-} **I**
b. H_2O **M**
c. O_2 **M**
d. Au^{3+} **I**

e. Ge **A**
f. O **A**
g. Ca^{2+} **I**
h. NH_3 **M**

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**Keith Johnson, Sue Adamson, Gareth
Williams**



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