

Chapter 9 P&T Questions — Part B (Observations and Examples for Part A, for Chapter 9)

Given the reaction $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ + 50 kJ, calculate the enthalpy

- A. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- B. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- C. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- D. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$



Given the following sequence of experiments, predict the final result.

- A. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- B. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- C. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- D. $\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$

Answers Chapter 1mini Pat

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