

- (a) Substance C-13 is an organic compound. Perform the experiments given below. Record the changes taking place at each step of the experiment.

Note the smell of the substance formed, the colour of the substance obtained, the colour of the precipitate produced, changes on heating and cooling and any other observations you may have. State the identity of the compound on the basis of the experiments and observational changes.

Substance C-13

PROCEDURE:

- (i) Take 2 ml of C-13 in a test tube. To this, add 1 ml of Tollen's reagent. Warm the contents in a water bath.
- (ii) Take 2 ml of C-13 in a test tube and add 1 ml of freshly prepared pyrogallol solution. Shake the contents. Add 2 ml of concentrated hydrochloric acid and warm the contents in a water bath.
- (iii) Take 2 ml of C-13 in a test tube and add a few crystals of resorcinol, shake the contents. Slowly add 1 ml of concentrated sulphuric acid along the sides of the test tube.

- (b) Substance C-14 is an unknown sample of either carbohydrate or protein. Carry out the following experiments and record all your observations. State the identity of the compound as carbohydrate or protein on the basis of the experiments and observational changes.

Substance C-14

PROCEDURE:

Take the sample C-14 in a test tube. Dissolve it in 10 ml of distilled water in order to obtain saturated solution. Divide the solution into three parts.

- (i) To the first part of C-14, add 2 drops of alcoholic α -naphthol solution followed by 1 ml of concentrated H_2SO_4 carefully by the side of the test tube.
- (ii) To the second part of C-14, add 1 ml of lead acetate solution, heat to boil. Now, add 5 ml of ammonium hydroxide solution and heat to boil again.
- (iii) To the third part of C-14, add 1 ml of copper sulphate solution, followed by 3 ml of sodium hydroxide solution.

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S Ben Porath



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