

## CHAPTER 6

## STUDY GUIDE

## Ionic Compounds and Metals

## Lesson 6.1 Ion Formation

In your textbook, read about chemical bonds and formation of ions.

Use each of the terms below just once to complete the passage.

chemical bond	electrons	energy level	ions	noble gases
nucleus	octet	pseudo-noble gas formations	valence	

The force that holds two atoms together is called a(n) (1) Chemical Bond. Such an attachment may form by the attraction of the positively charged (2) Nucleus of one atom for the negatively charged (3) Electron of another atom, or by the attraction of charged atoms, which are called (4) Ions. The attractions may also involve (5) Valence electrons, which are the electrons in the outermost (6) Energy level. The (7) Noble Gas are a family of elements that have very little tendency to react. Most of these elements have a set of eight outermost electrons, which is called a stable (8) Octet. The relatively stable electron structures developed by loss of electrons in certain elements of groups 3, 4, 13, and 14 are called (9) Pseudo-Noble gas formations.

For each statement below, write *true* or *false*.

- False 10. A positively charged ion is called an anion.
- True 11. Elements in group 1 lose their one valence electron, forming an ion with a 1+ charge.
- False 12. Elements tend to react so that they acquire the electron structure of a halogen.
- True 13. A sodium atom tends to lose one electron when it reacts.
- True 14. The electron structure of a zinc ion ( $\text{Zn}^{2+}$ ) is an example of a pseudo-noble gas formation.
- False 15. A  $\text{Cl}^-$  ion is an example of a cation.
- True 16. The ending *-ide* is used to designate an anion.
- False 17. Nonmetals form a stable outer electron configuration by losing electrons and becoming anions.

# Answer Study Guide Ionic Compounds And Metals

**Kenneth W. Raymond**



## **Answer Study Guide Ionic Compounds And Metals:**

*Hazmat Chemistry Study Guide (Second Edition)* Jill Meryl Levy, 2011      **Chemistry, Student Study Guide** James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane  $\text{CH}_4$  from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

**Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1992      **Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today**

Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application      **General, Organic, and**

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supplement includes for each chapter a brief overview activities and practice problems to reinforce skills and a practice test The answers section includes answers for all odd numbered end of chapter exercises      *Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry* Susan McMurry, 1996 This internationally acclaimed

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jeder von 7 bis 70 jahren kann windsurfen lernen step by step völlig ungefährlich mit viel spannung und spaß das jedenfalls behaupten die meisten wassersportlehrer und sie haben recht für alle die sicher schwimmen können ist windsurfen kinderleicht zu erlernen vorausgesetzt die anleitung stimmt ab wann geht s los

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egal ob sie in einer surfschule oder auf eigene faust surfen lernen möchten das fachwissen aus diesem buch sorgt für mehr spaß auf dem board entspannt über das wasser gleiten wellen abreiten freestyle tricks und sprünge mit dem richtigen know how ist im windsurfen alles möglich mehr

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in schweizer surfschulen lässt sich der wassersport lernen windsurfen wird mit einem surfbrett ausgeübt dabei dient ein dreh und kippbare segel der fortbewegung und der steuerung erfahrene surfer nutzen das segel aber auch um beeindruckende manöver zu erproben doch wie lernt man windsurfen am besten auf eigene faust oder in einem

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