

Name: Arturo CarbonellDate: 7/15/21

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? They should share Markers to have a full set.

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? The should share markers to have a full set

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? 7 Valence electrons
2. Click **Pause** (). Drag a valence electron from the left atom to the right atom. Click **Play**. What happens? A shell is added that connects the two atoms.
3. Click **Pause**, drag a valence electron from the right atom to the left, and then click **Play**. What happens now? An electron is added to the shell that is connecting the two atoms.

Answer Key To Student Exploration Covalent Bonds

Bruce Robertson



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Table of Contents Answer Key To Student Exploration Covalent Bonds

1. Understanding the eBook Answer Key To Student Exploration Covalent Bonds
 - The Rise of Digital Reading Answer Key To Student Exploration Covalent Bonds
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Student Exploration Covalent Bonds
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Student Exploration Covalent Bonds
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Student Exploration Covalent Bonds
 - Personalized Recommendations
 - Answer Key To Student Exploration Covalent Bonds User Reviews and Ratings
 - Answer Key To Student Exploration Covalent Bonds and Bestseller Lists
5. Accessing Answer Key To Student Exploration Covalent Bonds Free and Paid eBooks
 - Answer Key To Student Exploration Covalent Bonds Public Domain eBooks
 - Answer Key To Student Exploration Covalent Bonds eBook Subscription Services
 - Answer Key To Student Exploration Covalent Bonds Budget-Friendly Options
6. Navigating Answer Key To Student Exploration Covalent Bonds eBook Formats

- ePub, PDF, MOBI, and More
 - Answer Key To Student Exploration Covalent Bonds Compatibility with Devices
 - Answer Key To Student Exploration Covalent Bonds Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To Student Exploration Covalent Bonds
 - Highlighting and Note-Taking Answer Key To Student Exploration Covalent Bonds
 - Interactive Elements Answer Key To Student Exploration Covalent Bonds
 8. Staying Engaged with Answer Key To Student Exploration Covalent Bonds
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Student Exploration Covalent Bonds
 9. Balancing eBooks and Physical Books Answer Key To Student Exploration Covalent Bonds
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Student Exploration Covalent Bonds
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Answer Key To Student Exploration Covalent Bonds
 - Setting Reading Goals Answer Key To Student Exploration Covalent Bonds
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Answer Key To Student Exploration Covalent Bonds
 - Fact-Checking eBook Content of Answer Key To Student Exploration Covalent Bonds
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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