

Section 1: Introduction

Because of the high concentration of the product, the reaction mixture was extremely viscous. The reaction mixture with the addition of the starting ether dissolved in a methanol/water bath had an exotherm that was about 20°C. The heat developed was greater than the heat of the reaction, so the reaction was exothermic. The heat and the reaction heat will be later added together to determine the overall heat of the reaction that produces these products. The reaction mixture was then heated to 100°C. and kept at 100°C. for 1 hour. The reaction mixture was then cooled to room temperature and the product was isolated by vacuum distillation. The product was isolated in 90% yield. The product was isolated in 90% yield.

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• **Chlorophyll a fluorescence** is a measure of the amount of energy available for photosynthesis.

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