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Applebees Calorie Guide

Steven G. Aldana



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web oct 9 2023 second law of thermodynamics the entropy of an isolated system not in equilibrium will tend to increase over time approaching a maximum value at equilibrium $\Delta S = \Delta q / T$ $\Delta S_{\text{total}} = \Delta S_{\text{system}} + \Delta S_{\text{surrounding}}$ the second law of thermodynamics introduces the concept of entropy