

Multiple Choice:

Selecting the alternative that constitutes the statement or answers the question. What is most commonly observed with water?

1. How many atoms of hydrogen (protons) is a molecule of liquid water made of?
 a. 1
 b. 2
 c. 3
 d. 4

2. At standard $T_m = 100^\circ\text{C}$, what factor describes hydrogen bonding?
 a. 100
 b. 10
 c. 1
 d. 0.1

3. The strongest interaction in the liquid state:
 a. Van der Waals
 b. Hydrogen bonding
 c. Ionic interaction
 d. Covalent bonding

4. Pure water at $T_m = 100^\circ\text{C}$ is $100\% \text{ H}_2\text{O}$. How many atoms of hydrogen (protons) are contained in that as a molecule?
 a. 100
 b. 1000
 c. 10000
 d. 100000

5. What is the state of the water molecules when a molecular protein is dissolved in water?
 a. Dissolved molecules
 b. Hydrogen bonding
 c. Protein film
 d. Protein molecules

6. What is the basic nature of hydrogen?
 a. 100% hydrogen
 b. 100% oxygen
 c. 100% water
 d. 100% gas

7. A molecule's empirical formula is H_2O . At the molecular level, it is given as H_2O . What is the molecular formula?
 a. H_2O
 b. H_2O_2
 c. H_2O_3
 d. H_2O_4

8. Pure water at $T_m = 100^\circ\text{C}$ is $100\% \text{ H}_2\text{O}$. How many (protons of a molecule) are contained in that as a molecule?
 a. 100
 b. 1000
 c. 10000
 d. 100000

9. At standard $T_m = 100^\circ\text{C}$, what factor describes hydrogen bonding?
 a. 100
 b. 10
 c. 1
 d. 0.1

10. What is the state of the water molecules when a molecular protein is dissolved in water?
 a. Dissolved molecules
 b. Hydrogen bonding
 c. Protein film
 d. Protein molecules

11. The molecular formula of water is H_2O . At the molecular level, it is given as H_2O . What is the molecular formula?

Multiple Choice Stoichiometry Test

Kathleen Armour

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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Cambridge Checkpoints VCE Chemistry Units 3 and 4 2013 Roger Slade, Maureen Slade, 2012-12-14 Cambridge Checkpoints VCE are updated regularly to provide you with the most up to date exam preparation available **Roadmap to the Regents** Sasha Alcott, 2003 If Students Need to Know It It's in This Book This book develops the chemistry skills of high school students It builds skills that will help them succeed in school and on the New York Regents Exams Why The Princeton Review We have more than twenty years of experience helping students master the skills needed to excel on standardized tests Each year we help more than 2 million students score higher and earn better grades We Know the New York Regents Exams Our experts at The Princeton Review have analyzed the New York Regents Exams and this book provides the most up to date thoroughly researched practice possible We break down the test into individual skills to familiarize students with the test's structure while increasing their overall skill level We Get Results We know what it takes to succeed in the classroom and on tests This book includes strategies that are proven to improve student performance We provide a breakdown of the skills based on New York standards and objectives hundreds of practice questions organized by skill two complete practice New York Regents Exams in Physical Setting Chemistry **CLEP Success** Peterson's, 2007-05-04 Offers advice on improving scores on the CLEP and includes reviews for five CLEP subjects and five full length practice tests with explanatory answers **Master the CLEP** Peterson's, 2012-04-30 Peterson's Master the CLEP is the essential prep guide for non traditional students who are looking to begin a college career and for current students who want to save on tuition costs by testing out of certain courses When you use this guide you will get a basic overview of the CLEP a look at the general examinations and the vast number of subject specific exams tips for registering and preparing for the CLEP and a guide to interpreting your test scores This is followed by extensive reviews of the five CLEP general examinations with pre and post tests and subject matter overviews that will help you pinpoint your strengths and weaknesses The Appendix will go into detail about a number of CLEP subject exams Tests Richard C. Sweetland, Daniel J. Keyser, 1983 Descriptions of over 3000 tests in English intended as a guide for psychologists educators and other personnel who need test information to meet their assessment needs Topical arrangement under 3 main sections of psychology education and business Each entry gives test name associated personal names intended age group purpose description time range scoring cost and publisher Indexes by test titles authors publishers visually impaired tests and scoring services Teaching Chemistry Around the World Björn Risch, 2010 As teachers we often tend to expect other countries to teach chemistry in much the same way as we do but educational systems differ widely At Bielefeld University we started a project to analyse the approach to chemical education in different countries from all over the world Teaching Chemistry around the World 25 countries have participated in the project The resulting country studies are presented in this book This book may be seen as a contribution to make the structure of chemistry teaching in numerous countries more transparent and to facilitate communication between these

countries Especially in the case of the school subject chemistry which is very unpopular on the one hand and occupies an exceptional position on the other hand due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving we have to learn from each others educational systems

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