

Appendix 1

Abbreviations and symbols for quantities and units

Unavoidably, some symbols are used for more than one thing. The meaning is usually clear from the context, but always check that you understand what a symbol is representing. Note that symbols representing *physical quantities* are printed in *italics*. Symbols for units are *not* in italics. The reference (in brackets) given for each entry is to chapters where the abbreviation or symbol is used.

<i>A</i>	ampere (unit of electrical current) (1, 16)	$^{\circ}\text{C}$	degrees Celsius (1)
<i>A</i>	absorbance (10, 11, 28)	C_p	specific heat capacity (13)
<i>A</i>	area (8, 13, Maths Toolkit MT5)	<i>c</i>	concentration of a solution (1)
<i>A</i>	frequency factor (in Arrhenius equation) (9)	<i>c</i>	speed of light (2, 16)
<i>A</i>	Madelung constant (6)	<i>c</i>	root mean square speed (of molecules) (8)
<i>A_r</i>	relative atomic mass (1)	<i>c</i>	mean speed (of molecules) (8)
\AA	angstrom (non-SI unit used for bond lengths) (1)	CFSE	crystal field stabilization energy (28)
<i>a</i>	acceleration (1, 8)	CI	chemical ionization (in mass spectrometry) (12)
<i>a</i>	thermodynamic activity (14, 15, 16, 17)	CIP	Cahn-Ingold-Prelog (sequence rules for <i>R/S</i> nomenclature of chiral centres) (18)
<i>a</i>	van der Waals constant for gases (8)	Chr	carboxybenzyl group (24)
<i>a₀</i>	Bohr radius (3)	cm	centimetre (unit of length) (1)
AAS	atomic absorption spectrometry (11)	cm^3	cubic centimetre (unit of volume) (1)
AES	atomic emission spectrometry (11)	cm^{-1}	reciprocal centimetre (unit of wavenumber) (10, 12)
AO	atomic orbital (4)		
Ac	acetyl (2)	D	debye (non-SI unit of electric dipole moment) (5, 17)
Ar	general aryl group (2)	<i>D</i>	bond dissociation enthalpy (4, 13)
[acac] [−]	pentane-2,4-dionato (acetylacetonato) (28)	\bar{D}	mean bond enthalpy (13)
atm	atmosphere (non-SI unit of pressure) (1, 8)	<i>D</i>	centrifugal distortion constant (in rotational spectroscopy) (10)
<i>B</i>	magnetic field strength (10)	<i>d</i>	distance (1)
<i>B</i>	rotational constant (10)	<i>d</i>	doublet (in NMR spectrum) (12)
<i>b</i>	van der Waals constant for gases (8)	DBU	1,8-diazabicyclo[5.4.0]undec-7-ene (19)
Bn	benzyl (2)	DMAP	4-(dimethylamino)pyridine (19)
Boc	<i>tert</i> -butoxycarbonyl group (24)	DMF	<i>N,N</i> -dimethylformamide (2, 20)
Bz	benzoyl (2)	DMSO	dimethyl sulfoxide (20)
bar	bar (unit of pressure) (1, 8)	Da	dalton (atomic mass unit) (1)
bipy	2,2'-bipyridine (28)	dien	1,4,7-triazasheptane (diethylenetriamine) (28)
bp	bonding pair (5)	dm^3	cubic decimetre (unit of volume) (1)
b.p.	normal boiling point (usually quoted in $^{\circ}\text{C}$) (1)		
br	broad (IR absorption) (12)	E	electrophile (19)
C	coulomb (unit of electric charge) (1, 16)	<i>E</i>	energy (3, 16)
<i>C</i>	Curie constant (in Curie's law for magnetic susceptibility) (28)	E_{kin}	kinetic energy (1, 10)
C_p	molar heat capacity at constant pressure (13, 14)	E_{pot}	potential energy (1, 10, 17)
C_v	molar heat capacity at constant volume (13, 14)		

Appendix D Chemistry Answers

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Appendix D Chemistry Answers:

Chemistry John Olmsted, Greg Williams, Robert C. Burk, 2020 Chemistry 4th Edition is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers and distinguish this text from other offerings It more accurately reflects the curriculum of most Canadian institutions Chemistry is sufficiently rigorous while engaging and retaining student interest through its accessible language and clear problem solving program without an excess of material and redundancy

Chemistry Trace Jordan, Neville R. Kallenbach, 2017 Chemistry The Molecules of Life offers chemical insights within the context of health pharmaceuticals and the function of biological molecules The contextualized presentation of topics gives students a broad introduction to chemistry and helps them to see the relevance of chemistry to their personal lives

Chemistry James N. Spencer, George M. Bodner, Lyman H. Rickard, 1999 CD ROM contains study tool illustrations photos definitions text interactive notebook drawing tools

Fiscal Year 1987

Department of Energy Authorization: Basic research programs Fusion Advisory Panel (U.S.), 1986 Chemistry, Man, and Society, 1983

General Organic and Biological Chemistry Kenneth W. Raymond, 2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features

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Introduction to Chemistry Martha J. Gilleland, Rebecca Williams, 1986 **Report of the Royal Commission Appointed to Inquire Into the Water Supply of the Metropolis: Appendices to minutes of evidence** Great Britain. Royal Commission on the Supply of Water to the Metropolis, 1893 **Educational Times** , 1896 Annual Report for the Year (with Appendices). Great Britain Dept. of Science and Art, 1885 *Chemistry and Society* Mark Martin Jones, 1987 Education Outlook , 1892 Chemistry Stanley R. Radcliff, Marjorie H. Navidi, 1990 Physical Chemistry Thomas Engel, 2006 *Chemistry, Man, and Society* Mark Martin Jones, 1980 Publisher and Bookseller , 1893 Vols for 1871 76 1913 14 include an extra number The Christmas bookseller separately paged and not included in the consecutive numbering of the regular series *The Bookseller and the Stationery Trades' Journal* , 1893 Official organ of the book trade of the United Kingdom **Chemistry in the Modern World** Frank L. Wiseman, 1985

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