

Valence Electrons

Review

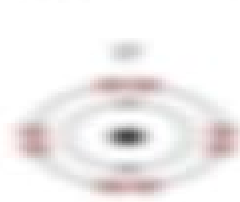
1) How many electrons are present in the following elements?

- ☒ Carbon 6 ☒ Chlorine 17 ☒ Sodium 11 ☒ Calcium 20
☒ Tungsten 74 ☒ Iodine 53 ☒ Titanium 22 ☒ Tin 50

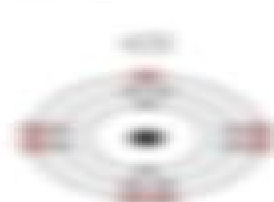
2) Using the periodic table, determine the number of valence electrons for the elements listed below.

- ☒ Copper 1 ☒ Boron 3 ☒ Hydrogen 1 ☒ Aluminum 3
☒ Barium 2 ☒ Sulfur 6 ☒ Carbon 4 ☒ Phosphorus 5

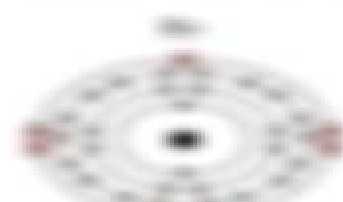
3) Circle the valence electrons only. What is the atomic number?



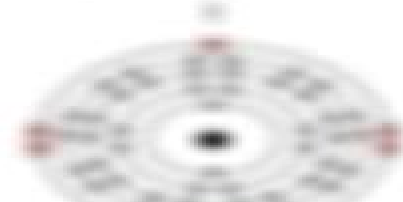
Atomic number(Z) = 3



Atomic number(Z) = 17



Atomic number(Z) = 50



Atomic number(Z) = 92

4) Fill in the table with correct values.

Element	Ion symbol	Number of Protons	Number of Electrons	Charge	Ion type
Phosphorus	P	15	15	0	Neutral
Sulfur	S	16	16	0	Neutral
Sulfur	S ²⁻	16	18	-2	Anion
Strontium	Str ²⁺	38	36	+2	Cation
Calcium	Ca ²⁺	20	18	+2	Cation

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