

Conjugate Acid-Base Pairs

Fill in the missing information in the two tables about conjugate acids and bases.

Acid	Base	Conjugate Acid	Conjugate Base	Equation
HNO_3	H_2O			$\text{HNO}_3 + \text{H}_2\text{O} \rightarrow \text{NO}_3^- + \text{H}_3\text{O}^+$
H_2O	F^-	HF	OH^-	
				$\text{NH}_3 + \text{HCN} \rightarrow \text{NH}_4^+ + \text{CN}^-$
		H_2O	ClO_3^-	
HSO_4^-	PO_4^{3-}			
				$\text{H}_2\text{O} + \text{S}^{2-} \rightarrow \text{HS}^- + \text{OH}^-$
HCO_3H	OH^-			

Acid	Conjugate Base	Equation
H_2PO_4^-		
	F^-	
	NO_3^-	
H_2PO_4^-		
H_2O		
	SO_4^{2-}	
HPO_4^{2-}		

Answers For Conjugate Acid Base Pairs

**Alfred N. Martin,James
Swarbrick,Arthur Cammarata**

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Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter. Key revisions in this edition focus on three areas. The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry. Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering, biological, pharmaceutical and medical sciences. Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text. When students know what they know they are better able to learn and incorporate the material. Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment, answer specific responses and additional practice resources. The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge. Problems are arranged in an intuitive confidence building order.

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