

Conjugate Acid-Base Pairs

Fill in the missing information in the two tables about conjugate acids and bases.

Acid	Base	Conjugate Acid	Conjugate Base	Equation
HNO_2	H_2O			$\text{HNO}_2 + \text{H}_2\text{O} \rightarrow \text{NO}_2^- + \text{H}_3\text{O}^+$
H_2O	F^-	HF	OH^-	
				$\text{NH}_3 + \text{HCN} \rightarrow \text{NH}_4^+ + \text{CN}^-$
		H_2O	ClO_3^-	
HSO_4^-	PO_4^{3-}			
				$\text{H}_2\text{O} + \text{S}^{2-} \rightarrow \text{HS}^- + \text{OH}^-$
HCO_2H	OH^-			

Acid	Conjugate Base	Equation
H_3PO_4		
	F^-	
	NO_3^-	
H_2PO_4^-		
H_2O		
	SO_4^{2-}	
HPO_4^{2-}		

Answers For Conjugate Acid Base Pairs

**Alfred N. Martin, James
Swarbrick, Arthur Cammarata**



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