



balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay

electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE (naming compounds, writing formulas)

Write the formula for each of the following compounds:

a. Sulfur hexafluoride SF_6

b. Lithium nitride Li_3N

c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$

d. Tin (II) chloride SnCl_2

e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)

f. Mercury (I) chloride Hg_2Cl_2

g. Potassium bromate KBrO_3

h. Hydrosulfuric acid $\text{H}_2\text{S(OH)}_2$

i. Chloric acid HClO_3

j. Sulfurous acid H_2SO_3

Name each of the following compounds.

a. CuSO_4 Copper (II) sulfate

b. AlF_3 aluminum fluoride

c. HI hydrogen iodide

d. NO nitrogen monoxide

e. H_2Se hydrogen selenide

f. HNO_3 nitric acid

g. HNO_2 nitrous acid

h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it takes for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{r} 63 \times .692 = 43.596 \rightarrow 43 \\ 65 \times .308 = 20.02 \rightarrow 20 \\ \hline 63.62 \rightarrow 64 \end{array}$$

Answers Chapter 9 Mixed Review Modern Chemistry

**Holt Rinehart & Winston, Holt,
Rinehart and Winston Staff**

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Modern Chemistry Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 *A Complete Introduction to Modern NMR Spectroscopy* Roger S. Macomber,1997-12-23 Clear accessible coverage of modern NMR spectroscopy for students and professionals in many fields of science Nuclear magnetic resonance NMR spectroscopy has made quantum leaps in the last decade becoming a staple tool in such divergent fields as chemistry physics materials science biology and medicine That is why it is essential that scientists working in these areas be fully conversant with current NMR theory and practice This down to basics text offers a comprehensive up to date treatment of the fundamentals of NMR spectroscopy Using a straightforward approach that develops all concepts from a rudimentary level without using heavy mathematics it gives readers the knowledge they need to solve any molecular structure problem from a complete set of NMR data Topics are illustrated throughout with hundreds of figures and actual spectra Chapter end summaries and review problems with answers are included to help reinforce and test understanding of key material From NMR studies of biologically important molecules to magnetic resonance imaging this book serves as an excellent all around primer on NMR spectroscopic analysis

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aqueous solutions of proteins or systems that are far from normal conditions e g critical or near critical mixtures Second the conventional thermodynamic methods are not sufficient for the accurate treatment of these mixtures Last but not least these systems are of interest for the pharmaceutical biomedical and related industries In order to meet the thermodynamic challenges involved in these complex mixtures we employed a variety of traditional methods but also new methods such as the fluctuation theory of Kirkwood and Buff and ab initio quantum mechanical techniques The Kirkwood Buff KB theory is a rigorous formalism which is free of any of the approximations usually used in the thermodynamic treatment of multicomponent systems This theory appears to be very fruitful when applied to the above mentioned difficult systems

Applied Mechanics Reviews ,1989 **The Electrician** ,1897 **Curriculum Review** ,1982 **The Electrical Journal** ,1897 *The*

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Modern Powder

Diffraction David L. Bish,1989 Principles of powder diffraction Scattering by points Calculation of the three dimensional diffraction pattern Instrumentation Measurement of X ray powder diffraction patterns camera methods Effect of instrument parameters X ray detectors Techniques for the selection of monochromatic radiation Resolutions versus intensity Experimental procedures Factors influencing d Spacing accuracy Factors influencing intensity accuracy Automation of qualitative phase analysis Use of standards Sample preparation for X ray diffraction Randomly oriented powder preparation Preparing oriented clay mineral aggregates Quantitative analysis Absorption diffraction method Exemple use of the reference intensity ratio or chung method Diffraction by small and disordered crystals Mixed layered micas and clay minerals Diffraction from glass Computer analysis of diffraction data Profile fitting of powder diffraction patterns Rietveld refinement of crystal structures using powder X ray diffraction data Synchrotron powder diffraction Neutron powder diffraction Reaction neutron sources Special sample environments in neutron diffraction

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