

CHEMISTRY PAPER-II

OBJECTIVE

TIME ALLOWED: 20 Minutes

MAXIMUM MARKS: 17

Note: You have four choices for each objective type question as A, B, C and D. The choice which you think is correct, fill that bubble in front of that question number, on bubble sheet. Use marker or pen to fill the bubbles. Cutting or filling two or more bubbles will result in zero mark in that question. No credit will be awarded in case BUBBLES are not filled. Do not solve question on this sheet of OBJECTIVE PAPER.

Q.No.1

- (1) Which one of the following does not belong to alkaline earth metals?
(A) *Be* (B) *Ra* (C) *Ba* (D) *Rn*
- (2) Which element forms an ion with charge +3?
(A) *Be* (B) *Al* (C) *C* (D) *Si*
- (3) Oxidation of *NO* in air produces:
(A) *N₂O* (B) *N₂O₃* (C) *N₂O₄* (D) *N₂O₅*
- (4) Chlorine heptaoxide (*Cl₂O₇*) reacts with water to form:
(A) Hypochlorous acid (B) Chloric acid (C) Perchloric acid (D) Chlorine and oxygen
- (5) The anhydride of *HClO₄* is:
(A) *ClO₃* (B) *ClO₂* (C) *Cl₂O₃* (D) *Cl₂O₇*
- (6) Which of the following is a non-typical transition element?
(A) *Cr* (B) *Mn* (C) *Zn* (D) *Fe*
- (7) Ethers show the phenomenon of:
(A) Position isomerism (B) Functional group isomerism (C) Metamerism (D) Cis-trans isomerism
- (8) Preparation of vegetable ghee involves:
(A) Halogenation (B) Hydrogenation (C) Hydroxylation (D) Dehydrogenation
- (9) Which of the following acid can be used as a catalyst in Friedal-Crafts reactions?
(A) *AlCl₃* (B) *HNO₃* (C) *BeCl₂* (D) *NaCl*
- (10) For which mechanisms, the first step involved is the same?
(A) *E₁* and *E₂* (B) *E₂* and *S_N2* (C) *S_N1* and *E₂* (D) *E₁* and *S_N1*
- (11) Which compound is more soluble in water?
(A) *C₂H₅OH* (B) *C₆H₅OH* (C) *CH₃COCH₃* (D) *n*-Hexanol
- (12) Which of the following will have the highest boiling point?
(A) Methanal (B) Ethanal (C) Propanal (D) 2-Hexanone
- (13) Which of the following reagents will react with both aldehydes and ketones?
(A) Grignard's reagent (B) Tollen's reagent (C) Fehling's reagent (D) Benedict's reagent
- (14) Which reagent is used to reduce a carboxylic group to an alcohol?
(A) *H₂/Ni* (B) *H₂/Pt* (C) *NaBH₄* (D) *LiAlH₄*
- (15) Alkanenitriles can be prepared by treating alkyl halide with:
(A) Alcoholic *KOH* (B) Alcoholic *KCN* (C) Aqueous *KOH* (D) Aqueous *KNO₃*
- (16) Most concentrated solid nitrogen fertilizer is:
(A) *NH₃* (B) *(NH₄)₂HPO₄* (C) *(NH₂)₂CO* (D) *NH₄NO₃*
- (17) Mark the correct statement.
(A) *Na⁺* is smaller than *Na* atom (B) *Na⁺* is larger than *Na* atom
(C) *Cl⁻* is smaller than *Cl* atom (D) *Cl⁻* ion and *Cl* atom are equal in size

Answers Of Hsc Chemistry Mcqs 2015

Arshad Iqbal



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Cambridge Checkpoints HSC Chemistry 2017-19 Maureen Slade, Roger Slade, 2016-06-29 *Gold Ore Processing* Mike D. Adams, 2016-05-03 Gold Ore Processing Project Development and Operations Second Edition brings together all the technical aspects relevant to modern gold ore processing offering a practical perspective that is vital to the successful and responsible development operation and closure of any gold ore processing operation This completely updated edition features coverage of established newly implemented and emerging technologies updated case studies and additional topics including automated mineralogy and geometallurgy cyanide code compliance recovery of gold from e waste handling of gaseous emissions mercury and arsenic emerging non cyanide leaching systems hydro re mining water management solid liquid separation and treatment of challenging ores such as double refractory carbonaceous sulfides Outlining best practices in gold processing from a variety of perspectives Gold Ore Processing Project Development and Operations is a must have reference for anyone working in the gold industry including metallurgists geologists chemists mining engineers and many others Includes several new chapters presenting established newly implemented and emerging technologies in gold ore processing Covers all aspects of gold ore processing from feasibility and development stages through environmentally responsible operations to the rehabilitation stage Offers a mineralogy based approach to gold ore process flowsheet development that has application to multiple ore types *Emerging Energy Alternatives for Sustainable Environment* D. P. Singh, Richa Kothari, V. V. Tyagi, 2019-02-14 Sustainability of environment is an emerging global issue at present Unsustainable or deteriorating environment is a matter of concern as it has threatened the survival of living creatures Recently climate change has been a matter of great concern at a global platform owing to imbalances in natural environment Increasing population has increased the demand for energy which has ultimately put pressure on natural resources and caused a paradigm shift from resource generation to exploitation Emerging Energy Alternatives for Sustainable Environment aims to address the role of sustainable technologies in energy generation options for clean environment It covers a wide spectrum of energy generation approaches with an emphasis on five key topics i renewable energy sources and recent advances ii emerging green technologies for sustainable development iii assessment of biomass for sustainable bioenergy production iv solid waste management and its potential for energy generation and v solar energy applications storage system and heat transfer This book provides essential and comprehensive knowledge of green energy technologies with different aspects for engineers technocrats and researchers working in the industry universities and research institutions The book is also very useful for undergraduate and graduate students of science and engineering who are keen to know about the development of renewable energy products and their corresponding processes Please note This volume is Co published with The Energy and Resources Institute Press New Delhi Taylor Francis does not sell or distribute the Hardback in India Pakistan Nepal Bhutan Bangladesh and Sri Lanka *Who's who in America* ,2003 **Chemistry** Karen Enkelaar, Rosemary

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Course Revision New South Wales. Board of Studies, 1994

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