

Practice Exam 2

Ch 6 & 7 (Chaps. 1, 2, 3, 4, 5)

1. Which of the following is a physical change?
 (a) rusting of metal (b) fruits ripen (c) baking a cake
 (d) ice melts (e) milk turns sour

2. Which of the following is a homogeneous mixture?
 (a) water and sand (b) sand, alcohol, and sugar
 (c) oil and vinegar (d) milk of magnesium (e) air

3. Which of the following is a correct?
 (a) sodium, S (b) mercury, Mg (c) iron, K, Sn
 (d) potassium, P (e) gold, Au

4. What is the equivalent temperature for 105 Fahrenheit in Kelvin?
 (a) 41 (b) 314 (c) 378 (d) 457 (e) -32

5. A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?
 (a) 1.59 (b) 1.74 (c) 2.36 (d) 2.89 (e) 3.50 $d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$

6. How many significant figures are there in the measurement of 1.0056×10^7 ?
 (a) 1 (b) 2 (c) 3 (d) 4 (e) 5

7. How many significant figures are appropriate to show in the result after carrying out the operation below?

$$(13.7 + 0.027) + 8.221 \rightarrow \text{X, } \div \text{ (last s.f.) } +, - \text{ (last decimal)}$$

 (a) 1 (b) 2 (c) 3 (d) 4 (e) 5

8. How many protons, electrons, and neutrons are there in $^{120}\text{Ba}^{2+}$?
 (a) 56p, 56e, 130n (b) 56p, 54e, 130n (c) 56p, 58e, 74n
 (d) 56p, 54e, 74n (e) 54p, 56e, 74n

9. Which of the following pairs would have similar chemical and physical properties?
 (a) Na, Mg (b) H, Li (c) He, Ne (d) Cr, Ca (e) Cl, O

10. Which of the following is the molecular formula?
 (a) NaCl (b) CsF (c) HCl (d) AlBr₃ (e) NaCN

11. Which of the following is the empirical formula?
 (a) C₂H₆ (b) CH₃COOH (c) P₂O₅ (d) NH₄NO₃ (e) Ag₂CO₃

12. Which of the following is a correct match?
 (a) BaCl₂, barium dichloride (b) CS₂, carbon sulfide
 (c) CuF₂, copper (II) fluoride (d) HNO₃, nitrous acid
 (e) MgSO₄·7H₂O, magnesium sulfate heptahydrate

13. Which of the following pairs is a correct match?
 (a) Ca₃(PO₄)₂, calcium (III) phosphate
 (b) (NH₄)₂SO₄, ammonium sulfate
 (d) Cu(NO₃)₂, copper dinitrate (c) Mg₂O₃, magnesium oxide
 (e) H₂CO₃, carbon acid

Answers Key For Chemistry 1411

VM Jensen

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Table of Contents Answers Key For Chemistry 1411

1. Understanding the eBook Answers Key For Chemistry 1411
 - The Rise of Digital Reading Answers Key For Chemistry 1411
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers Key For Chemistry 1411
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers Key For Chemistry 1411
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Key For Chemistry 1411
 - Personalized Recommendations
 - Answers Key For Chemistry 1411 User Reviews and Ratings
 - Answers Key For Chemistry 1411 and Bestseller Lists
5. Accessing Answers Key For Chemistry 1411 Free and Paid eBooks
 - Answers Key For Chemistry 1411 Public Domain eBooks
 - Answers Key For Chemistry 1411 eBook Subscription Services
 - Answers Key For Chemistry 1411 Budget-Friendly Options

6. Navigating Answers Key For Chemistry 1411 eBook Formats
 - ePUB, PDF, MOBI, and More
 - Answers Key For Chemistry 1411 Compatibility with Devices
 - Answers Key For Chemistry 1411 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers Key For Chemistry 1411
 - Highlighting and Note-Taking Answers Key For Chemistry 1411
 - Interactive Elements Answers Key For Chemistry 1411
8. Staying Engaged with Answers Key For Chemistry 1411
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers Key For Chemistry 1411
9. Balancing eBooks and Physical Books Answers Key For Chemistry 1411
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers Key For Chemistry 1411
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answers Key For Chemistry 1411
 - Setting Reading Goals Answers Key For Chemistry 1411
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answers Key For Chemistry 1411
 - Fact-Checking eBook Content of Answers Key For Chemistry 1411
 - Distinguishing Credible Sources
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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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