

CHAPTER 1 REVIEW*Matter and Change***SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

1. a Technological development of a chemical product often
- (a) lags behind basic research on the same substance.
 - (b) does not involve chance discoveries.
 - (c) is driven by curiosity.
 - (d) is done for the sake of learning something new.

2. d The primary motivation behind basic research is to
- (a) develop new products.
 - (b) make money.
 - (c) understand an environmental problem.
 - (d) gain knowledge.

3. a Applied research is designed to
- (a) solve a particular problem.
 - (b) satisfy curiosity.
 - (c) gain knowledge.
 - (d) learn for the sake of learning.

4. b Chemistry is usually classified as
- (a) a biological science.
 - (b) a physical science.
 - (c) a social science.
 - (d) a computer science.

5. Define the six major branches of chemistry.

organic chemistry—the study of carbon-containing compounds

inorganic chemistry—the study of non-organic substances

physical chemistry—the study of properties of matter, changes that occur in matter, and the relationships between matter and energy

analytical chemistry—the identification of the composition of materials

biochemistry—the study of the chemistry of living things

theoretical chemistry—the use of mathematics and computers to design and predict the properties of new compounds

Answer To Chemistry Matter And Change

Leo J. Malone



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Chemistry Laurel Dingrando, McGraw-Hill Staff, Glencoe/McGraw-Hill, 2001-08-01 Physical and Chemical Changes (ENHANCED eBook) Edward P. Ortleb, Richard Cadice, 1993-09-01 This book presents a program of basic studies in physical and chemical changes of matter. The definition of matter is presented along with explanations of states and properties of matter. Topics include atoms, molecules, elements, compounds, mixtures, solutions, symbols, and formulas. Each of the twelve teaching units in this book is introduced by a color transparency, print books, or PowerPoint slide eBooks that emphasize the basic concept of the unit and presents questions for discussion. Reproducible student pages provide reinforcement and follow up activities. The teaching guide offers descriptions of the basic concepts to be presented, background information, suggestions for enrichment activities, and a complete answer key.

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Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Denniston Katherine, 2006 *Chemistry and You*, 1962

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General Chemistry Henry Fuller Holtzclaw,William Harrison Nebergall,William R. Robinson,1984 The British National Bibliography Arthur James Wells,2000 American Druggist and Pharmaceutical Record ,1895 **American Chemical Journal** Ira Remsen,Charles August Rouiller,1900 **The Medical Times and Gazette** ,1875 *Chemistry* Laurel Dingrando,2005 *Basic Chemistry* Steven S. Zumdahl,1999 For a full description see catalog entry for Zumdahl Introductory Chemistry A Foundation 4 e *College Chemistry* Henry Fuller Holtzclaw,William R. Robinson,William Harrison Nebergall,1984

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