
AQA a level physics

paper 1 100%

SOLUTIONS REVISED

2023/2024

What is meant by work function? (2) - ANSWER Minimum energy required by an electron to escape from (1) a metal surface (1)

What is meant by ionisation energy? (2) - ANSWER Minimum energy required to remove an electron from an atom (1) from the ground state (1)

Why do emitted electrons have a kinetic energy that varies up to a maximum value? (4) - ANSWER Energy of a photon is fixed/constant (1)
Energy is required for electrons to overcome work function ϕ (1)
Electrons deeper into/below the surface of the metal require more energy to overcome work function and therefore have less KE(1)
Maximum KE = Energy of photon - work function ϕ (1)

Explain how (bright ring) diagram is evidence of the wave-particle duality of electrons? - ANSWER Diffraction pattern maxima/minima (bright and dark areas) (1)
Diffraction is a property of waves- proves wave-behaviour of electrons (1)

What is meant by the ground state? - ANSWER The lowest energy state of an atom

Explain the difference between excitation and ionisation (2) - ANSWER An electron receives a discrete amount of energy for BOTH (1)
Excitation promotes an electron to a higher energy level
Ionisation is when an electron receives enough energy to leave the atom

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 - The Rise of Digital Reading Answer On Physics Paper 2 And 1 In 2014 Weac
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer On Physics Paper 2 And 1 In 2014 Weac
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer On Physics Paper 2 And 1 In 2014 Weac
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer On Physics Paper 2 And 1 In 2014 Weac
 - Personalized Recommendations
 - Answer On Physics Paper 2 And 1 In 2014 Weac User Reviews and Ratings
 - Answer On Physics Paper 2 And 1 In 2014 Weac and Bestseller Lists
5. Accessing Answer On Physics Paper 2 And 1 In 2014 Weac Free and Paid eBooks
 - Answer On Physics Paper 2 And 1 In 2014 Weac Public Domain eBooks
 - Answer On Physics Paper 2 And 1 In 2014 Weac eBook Subscription Services
 - Answer On Physics Paper 2 And 1 In 2014 Weac Budget-Friendly Options

6. Navigating Answer On Physics Paper 2 And 1 In 2014 Weac eBook Formats
 - ePUB, PDF, MOBI, and More
 - Answer On Physics Paper 2 And 1 In 2014 Weac Compatibility with Devices
 - Answer On Physics Paper 2 And 1 In 2014 Weac Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer On Physics Paper 2 And 1 In 2014 Weac
 - Highlighting and Note-Taking Answer On Physics Paper 2 And 1 In 2014 Weac
 - Interactive Elements Answer On Physics Paper 2 And 1 In 2014 Weac
8. Staying Engaged with Answer On Physics Paper 2 And 1 In 2014 Weac
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer On Physics Paper 2 And 1 In 2014 Weac
9. Balancing eBooks and Physical Books Answer On Physics Paper 2 And 1 In 2014 Weac
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer On Physics Paper 2 And 1 In 2014 Weac
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer On Physics Paper 2 And 1 In 2014 Weac
 - Setting Reading Goals Answer On Physics Paper 2 And 1 In 2014 Weac
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer On Physics Paper 2 And 1 In 2014 Weac
 - Fact-Checking eBook Content of Answer On Physics Paper 2 And 1 In 2014 Weac
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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