



Atoms and Elements

Name: _____

Date: _____

chemical	mixtures	two-letter	brassine	bonds
chemical	cannot	gold	nitrogen	mercury
Mg	cannot	sulfur	different	compound
element	oxygen	hydrogen	impure	different
name	fluorine	distilling	carbon	Compounds
nitrogen	metals	hydrogen	elements	molecules
bonded	hydrogen	one	solids	atoms
atom	element	physical	single-letter	lattices
different	helium	filtering	oxygen	element
chemical	atom			

There is a huge number of _____ chemicals in our world. Some of these chemicals that will be familiar to you are glass, wood, water, salt, sugar, petrol and plastic.

These chemicals are all made out of extremely small particles called _____, which are joined together in _____ (groups of atoms) or _____ (networks).

All of the thousands of different chemicals that exist are made up from a limited number of different kinds of atoms.

An _____ is a simple chemical made up of only _____ type of atom. Pure gold is an element because it contains _____ atoms only.

An element _____ be broken down into simpler substances. Water can be broken down into _____ and _____ gases, so it is not an _____. Liquid mercury _____ be broken down into simpler substances, so it is an _____.

Most of the elements are _____. Eleven are gases (_____, _____, _____, nitrogen, oxygen, fluorine, neon, chlorine, etc.) and just two are liquids (_____ and _____).

Most of the elements are _____. Only 20 or so are non-metals (carbon, _____, _____, neon, phosphorus, sulfur, chlorine, iodine, etc.).

Each element has been given a distinct _____. Often the name was given to the element by the scientist.

Atoms Elements Compounds Cloze Activity

Julia Burton

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Science Richard Moyer,2000 For grades 1-6 *The software catalog microcomputers* Menu (Firm) (Fort Collins, Colo.),1989 *The ETS Test Collection Catalog* Educational Testing Service. Test Collection,1993 The major source of information on the availability of standardized tests Wilson Library BulletinCovers commercially available standardized tests and hard to locate research instruments **Atoms, Molecules & Elements: What Are Molecules? Gr. 5-8** George Graybill,2015-10-01 This is the chapter slice What Are Molecules from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives **Atoms, Molecules & Elements: What Are Atoms? Gr. 5-8**

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activities in this book explain elementary concepts in the study of chemistry including matter elements of the earth's crust and compounds and mixtures General background information suggested activities questions for discussion and answers are included

Atom, Grades 6 - 12 Pat Ward, Barbara Ward, 1995-01-01 In this captivating classroom supplement students examine atoms the building blocks of nature Topics covered include matter atomic structure electrons Mendeleyev the periodic table elements compounds solutions mixtures and more Information is presented in fascinating passages and reinforced with a variety of activities A complete answer key is also included Mark Twain Media Publishing Company specializes in providing captivating supplemental books and decorative resources to complement middle and upper grade classrooms Designed by leading educators the product line covers a range of subjects including mathematics sciences language arts social studies history government fine arts and character Mark Twain Media also provides innovative classroom solutions for bulletin boards and interactive whiteboards Since 1977 Mark Twain Media has remained a reliable source for a wide variety of engaging classroom resources

Chemistry Activity Book Jenny Jacoby, 2020-10 *Elements, Compounds, and Mixtures* J. M. Patten, 1995 Explains the science of elements compounds and mixtures and includes photographs and a glossary

Atoms and Molecules Chris Woodford, Martin Clowes, 2012-07-15 Greek philosophers first hypothesized that matter was composed of atoms but the theory would not resurface again until the late 17th century The idea that atoms joined to form structures called molecules first appeared in the 19th century and helped explain why gases liquids and solids behave differently from one another In the 20th century subatomic particles were discovered electrons protons and neutrons and atomic structure was finally understood These breakthroughs led to the development of quantum theory and quantum mechanics This book details the inspiring and heroic discovery delving deeply into intriguing stories reviewing major scientific landmarks and introducing readers to the vivid men and women who helped discover and map the microscopic universe that is the atom Supplemental content includes an activity spread a substantial and highly detailed timeline and a list of key people with mini biographies

Atoms and Chemical Reactions Suzanne Slade, 2006-08-15 Introduces atoms molecules elements compounds and mixtures and how chemical reactions occur

Investigating the Natural World of Chemistry with Kids Michael J. Strauss, 2012-09 This unique book of real chemistry and science for children illustrates the nature of physical and chemical change using the very smallest parts of things atoms and molecules It encourages children ages 5-12 along with their parents or teachers to become active learners of science to discover meaning not only in the ideas and definitions of others but also and especially in their own world Chapters include Evaporating Condensing Dissolving Crystallizing Mixing Separating Melting Freezing and Reacting

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