

# Atomic Structure & Isotope

## Part A: Atomic Structure

1. Complete the following table.

| Element name | Atomic number | Mass number | Number of protons | Number of electrons | Number of neutrons | Element symbol |
|--------------|---------------|-------------|-------------------|---------------------|--------------------|----------------|
|              |               | 70          |                   |                     |                    |                |
|              | 48            |             |                   |                     |                    |                |
|              |               |             |                   | 80                  |                    |                |
|              |               |             | 35                |                     |                    |                |

2. Identify the neutral atom described by name and mass number (example, oxygen-16).

i) The atom with 2 neutrons and 1 proton is \_\_\_\_\_.

ii) The atom with 17 electrons and 18 neutrons is \_\_\_\_\_.

iii) The atom with 6 protons and 8 neutrons is \_\_\_\_\_.

## Part B: Isotope

1. Complete the following table.

| Symbol          | # of protons | # of neutrons | # of electrons | Symbol                     | # of protons | # of neutrons | # of electrons |
|-----------------|--------------|---------------|----------------|----------------------------|--------------|---------------|----------------|
| $^1\text{H}_1$  |              |               |                | $^{17}\text{O}_8$          |              |               |                |
| $^2\text{H}_1$  |              |               |                | $^{39}\text{K}_{19}$       |              |               |                |
| $^4\text{He}_2$ |              |               |                | $^{40}\text{Ca}_{20}$      |              |               |                |
| $^6\text{He}_2$ |              |               |                | $^{238}\text{U}_{92}$      |              |               |                |
| $^6\text{Li}_3$ |              |               |                | $^{244}\text{Pu}_{94}$     |              |               |                |
| $^7\text{Li}_3$ |              |               |                | $^{56}\text{Fe}_{26}^{3+}$ |              |               |                |

# Atoms And Isotopes Worksheet Answers

**George Graybill**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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