

Atomic Theory Practice Test

Name:

Date:

Block:

- _____ 1. If two atoms of Ca have a different number of electrons, which property (or properties) would be significantly different?
- Mass
 - Charge
 - Both A & B
 - Neither A nor B
- _____ 2. If two atoms of Na have a different number of neutrons which property (or properties) would be significantly different?
- Mass
 - Charge
 - Both A & B
 - Neither A nor B
- _____ 3. Which of the following ions will NOT have the same electron configuration as Ne?
- Na^+
 - Al^{3+}
 - Ar
 - O^{2-}
- _____ 4. Which of the following easily loses one outermost electron?
- Potassium
 - Beryllium
 - Iron
 - Bromine
- _____ 5. Which of the following would have the largest atomic radius?
- Seaborgium
 - Einsteinium
 - Indium
 - Thallium
- _____ 6. Which trend in the halogen family occurs with increasing atomic number?
- Ionization energies decrease
 - Atomic radii decrease
 - Electronegativities increase
 - Tendency to gain electrons increases
- _____ 7. Which of the following species would have 2 valence electrons?
- Boron
 - Fluorine
 - Oxygen ion
 - Vanadium (III) ion
- _____ 8. What sub-shell is especially stable when it is half-filled?
- s-subshell
 - p-subshell
 - d-subshell
 - f-subshell

Atomic Theory Practice Test

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