

# Atomic Structure

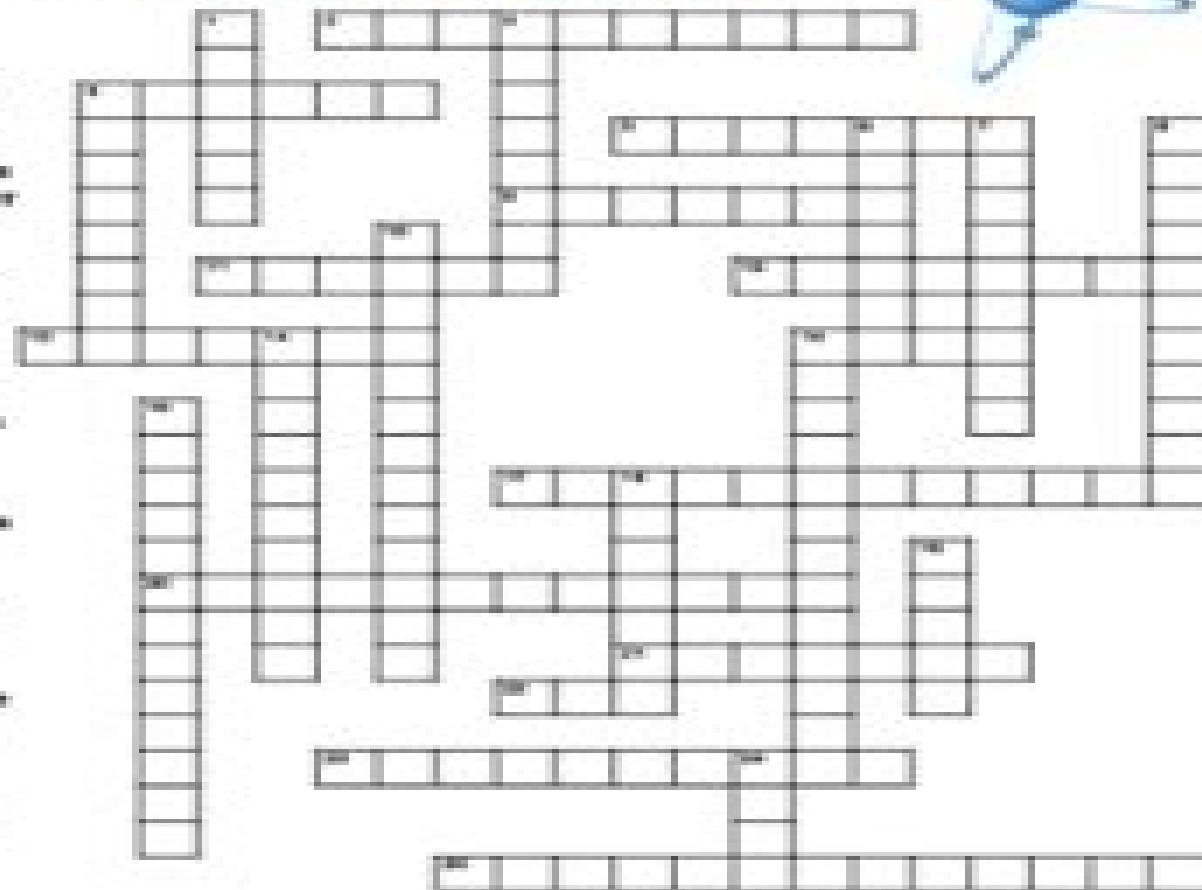


## Definitions

1. A form of energy.
2. The fundamental particle of visible light.
3. The content of an atom.
4. The term given to atoms that have the same number of protons but different numbers of neutrons.
5. More positive or negative an atom is.
6. A negatively charged subatomic particle.
7. A subatomic particle with no charge.
8. The smallest unit of an element.
9. Electrons emitted by radioactive decay.
10. The number of protons in an atom.
11. The region of space around the nucleus where an electron is likely to be found.
12. An atom that has a positive or negative charge due to the gain or loss of one or more electrons.
13. The total number of protons and neutrons in the nucleus of an atom.
14. Empty space around the nucleus of an atom where electrons travel.

## Identify

1. A subatomic particle with a positive charge.
2. The time for a radioactive substance to lose half of its radioactivity.
3. An increase in energy of matter.
4. A form of matter that contains the electrically neutral atoms that



## Thinking skills section

1. Smaller than or exceeding normal atoms.
2. The lowest possible energy state of an atom.
3. Changing of one element to another through radioactive decay.
4. New Zealand physicist who proposed the nuclear structure.
5. Two scientists and a zoologist

## Recall forgotten and predicted key radioactive decay

1. Electron and neutrino from an unstable atom.
2. English physicist who discovered electrons.
3. The basic unit smaller than.
4. The French physicist who made a small contribution to the understanding of atomic structure.

# Atomic Structure Crossword Worksheet Answers

**S Ashworth**

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