

Atomic Structure



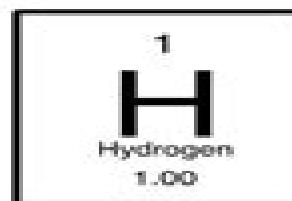
Atomic Number
Element Symbol
Element Name
Atomic Mass

The atomic number represents these two parts of the atom:
number of protons & number of electrons

You can calculate the number of neutrons by subtracting
Atomic Mass minus Atomic Number



Protons: 7
Electrons: 7
Neutrons: 7



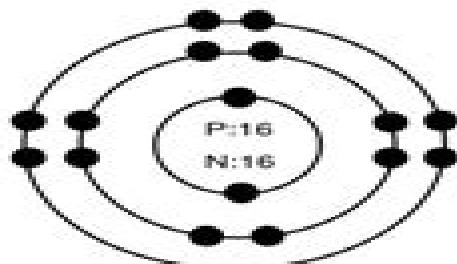
Protons: 1
Electrons: 1
Neutrons: 0



Protons: 8
Electrons: 8
Neutrons: 8

BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
 1. The first energy level can hold 2 electrons
 2. The second energy level can hold 8 electrons
 3. The third energy level can hold 8 electrons



Element: Sulfur

Number of Valence Electrons: 6

The electrons in the outermost energy level are called:

Valence Electrons

LEWIS DIAGRAMS

A Lewis diagram shows only the symbol with the Valence electrons



Atomic Answer Key

R Pring

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