

Name: \_\_\_\_\_

Period: \_\_\_\_\_

# Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:

number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



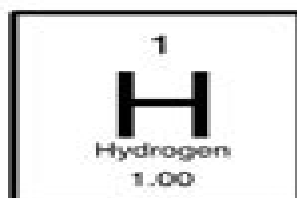
Protons:

7

Electrons:

7

Neutrons:

7

Protons:

1

Electrons:

1

Neutrons:

0

Protons:

8

Electrons:

8

Neutrons:

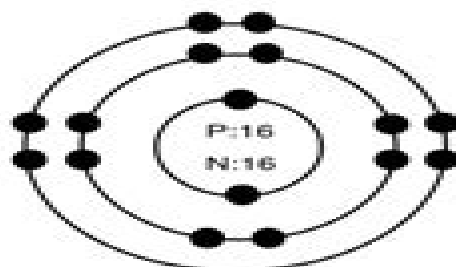
8

## BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
  1. The first energy level can hold 2 electrons
  2. The second energy level can hold 8 electrons
  3. The third energy level can hold 8 electrons

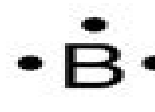
The electrons in the outermost energy level are called:

Valence Electrons

Element: SulfurNumber of Valence Electrons: 6

## LEWIS DIAGRAMS

A lewis diagram shows only the symbol with the Valence electrons



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**Clifford Lane**



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1. Understanding the eBook Atomic Structure Worksheet Key Matching
  - The Rise of Digital Reading Atomic Structure Worksheet Key Matching
  - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Worksheet Key Matching
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atomic Structure Worksheet Key Matching
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Worksheet Key Matching
  - Personalized Recommendations
  - Atomic Structure Worksheet Key Matching User Reviews and Ratings

- Atomic Structure Worksheet Key Matching and Bestseller Lists
- 5. Accessing Atomic Structure Worksheet Key Matching Free and Paid eBooks
  - Atomic Structure Worksheet Key Matching Public Domain eBooks
  - Atomic Structure Worksheet Key Matching eBook Subscription Services
  - Atomic Structure Worksheet Key Matching Budget-Friendly Options
- 6. Navigating Atomic Structure Worksheet Key Matching eBook Formats
  - ePub, PDF, MOBI, and More
  - Atomic Structure Worksheet Key Matching Compatibility with Devices
  - Atomic Structure Worksheet Key Matching Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Atomic Structure Worksheet Key Matching
  - Highlighting and Note-Taking Atomic Structure Worksheet Key Matching
  - Interactive Elements Atomic Structure Worksheet Key Matching
- 8. Staying Engaged with Atomic Structure Worksheet Key Matching
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Atomic Structure Worksheet Key Matching
- 9. Balancing eBooks and Physical Books Atomic Structure Worksheet Key Matching
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Atomic Structure Worksheet Key Matching
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Worksheet Key Matching
  - Setting Reading Goals Atomic Structure Worksheet Key Matching
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Worksheet Key Matching
  - Fact-Checking eBook Content of Atomic Structure Worksheet Key Matching
  - Distinguishing Credible Sources

13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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