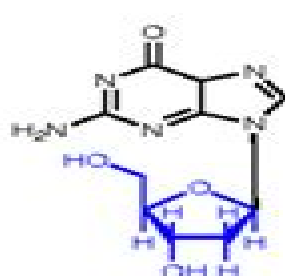


Chapter 6

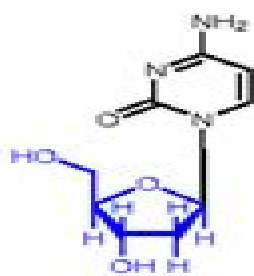
1) Proflavine contains a planar, heteroaromatic, tricyclic ring which could slip between the base pairs of DNA and interact with the base pairs through van der Waals and interactions. The amine groups are likely to be ionised at physiological pH and can form ionic bonds with the charged phosphate groups of the DNA sugar-phosphate backbone (see also section 9.1)

The drug is unlikely to show any selectivity between the DNA of bacteria and the DNA of human cells. Therefore, it cannot be used systemically.

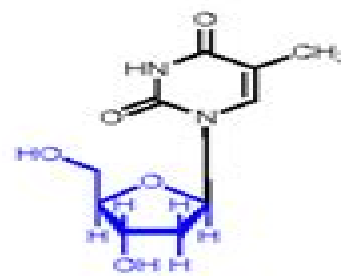
2) The nucleosides which are mimicked are deoxyguanosine, deoxycytidine, and deoxythymidine respectively (see also section 20.6.1)



Deoxyguanosine



Deoxycytidine



Deoxythymidine

3) It is possible to identify five CN fragments within the skeleton of adenine as shown below.



4) The table in appendix 2 can be used to answer this question. A change in Z is least likely to result in a change of amino acid. For example, CUU, CUC, CUA, and CUG all code for leucine.

5)
AGU to ACU

This changes serine to threonine. Both these molecules contain an alcohol functional group in their side chain, and so it is possible that the receptor would still be functional. The increased bulk of threonine might have an effect on the range of ligands that could bind.

Answers Patrick Medicinal Chemistry

AW Rasmussen



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