

## Student Exploration: Ionic Bonds

**Vocabulary:** chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

**Prior Knowledge Questions (Do these BEFORE using the Gizmo.)**

1. Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

**Answer:** Nate can give Clara one marker so they can both have 8 markers.

2. Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

**Answer:** Maggie can give one to both Fred and Florence.

### Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The **Ionic Bonds** Gizmo allows you to explore how **ionic bonds** form.

To begin, check that **Sodium (Na)** and **Chlorine (Cl)** are selected from the menus at right. Click **Play** (▶) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



1. Each atom consists of a central nucleus and several **shells** that contain electrons. The outermost electrons are called **valence electrons**. (Inner electrons are not shown.)

How many valence electrons does each atom have? Sodium: 1 Chlorine: 7

2. Click **Pause** (■). Elements can be classified as **metals** and **nonmetals**. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. **Electron affinity** is a measure of how tightly the valence electrons are held.

- A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

# Answers Key To Gizmo Ionic Bond

**SJ Ball**

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## **Table of Contents Answers Key To Gizmo Ionic Bond**

1. Understanding the eBook Answers Key To Gizmo Ionic Bond
  - The Rise of Digital Reading Answers Key To Gizmo Ionic Bond
  - Advantages of eBooks Over Traditional Books
2. Identifying Answers Key To Gizmo Ionic Bond
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answers Key To Gizmo Ionic Bond
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answers Key To Gizmo Ionic Bond
  - Personalized Recommendations
  - Answers Key To Gizmo Ionic Bond User Reviews and Ratings
  - Answers Key To Gizmo Ionic Bond and Bestseller Lists
5. Accessing Answers Key To Gizmo Ionic Bond Free and Paid eBooks
  - Answers Key To Gizmo Ionic Bond Public Domain eBooks

- Answers Key To Gizmo Ionic Bond eBook Subscription Services
- Answers Key To Gizmo Ionic Bond Budget-Friendly Options

6. Navigating Answers Key To Gizmo Ionic Bond eBook Formats

- ePub, PDF, MOBI, and More
- Answers Key To Gizmo Ionic Bond Compatibility with Devices
- Answers Key To Gizmo Ionic Bond Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Answers Key To Gizmo Ionic Bond
- Highlighting and Note-Taking Answers Key To Gizmo Ionic Bond
- Interactive Elements Answers Key To Gizmo Ionic Bond

8. Staying Engaged with Answers Key To Gizmo Ionic Bond

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Answers Key To Gizmo Ionic Bond

9. Balancing eBooks and Physical Books Answers Key To Gizmo Ionic Bond

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Answers Key To Gizmo Ionic Bond

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Answers Key To Gizmo Ionic Bond

- Setting Reading Goals Answers Key To Gizmo Ionic Bond
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Answers Key To Gizmo Ionic Bond

- Fact-Checking eBook Content of Answers Key To Gizmo Ionic Bond
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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