

13. (4 pts) What is an ionic compound?  
a. A compound that will dissolve in water  
b. Two nonmetal atoms bonded together  
c. Two metal atoms bonded together  
d. A metal and a nonmetal atom bonded together  
e. None of the above

14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?  
a. Ionic  
b. Covalent  
c. Both a and b  
d. Neither a nor b

15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?  
a. Ionic  
b. Covalent  
c. Both a and b  
d. Neither a nor b

16. (4 pts) How many total atoms are in this molecule:  $\text{C}_6\text{H}_{12}\text{OH}$ ?  
a. 3  
b. 4  
c. 19  
d. 20  
e. None of the above

17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is  $\text{C}_x\text{H}_{10}\text{Cl}_3$ , what must x be?  
a. 1  
b. 2  
c. 3  
d. 4  
e. None of the above

18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water?  $\text{C}_6\text{H}_5\text{ClF}$   $\text{C}_6\text{H}_5\text{CH}_3$   $\text{CS}_2$   $\text{BaSO}_4$   
a.  $\text{C}_6\text{H}_5\text{ClF}$   
b.  $\text{C}_6\text{H}_5\text{CH}_3$   
c.  $\text{CS}_2$   
d.  $\text{BaSO}_4$   
e.  $\text{CS}_2$  and  $\text{BaSO}_4$

# Answers For Chemistry Test Module 13

**Rachel S Tattersall**

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