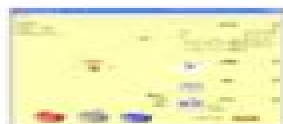


Atoms PhET Lab

Introduction: Atoms are the smallest things that retain the properties of matter we can observe. Atoms are made of three **subatomic** particles; protons, neutrons, and electrons.

- Protons have a mass of _____ amu and a charge of _____.
- Neutrons have a mass of _____ amu and a charge of _____.
- Electrons have a mass of nearly _____ amu and a charge of _____.

In this simulation, you will build atoms, subatomic particle by subatomic particle and observe the effect of adding more of each particle. When the subatomic particles in an atom change, an **ion**, **isotope** or different element will be created.

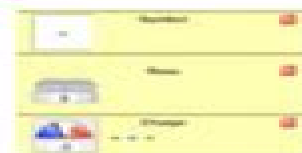


Build An Atom

Procedure 1: Play with the Sims → Chemistry → Build An Atom Run Sim

Begin by playing with the simulation for a while. Become familiar with the interface. What happens when you add protons, neutrons, or electrons? To start over, click Reset All.

Show the **symbol**, **atomic mass**, and **charge** by clicking on the Show.



Analysis Questions

1. Ions are atoms of the same element with different numbers of _____.
2. Isotopes are atoms of the same element with different numbers of _____.
3. Adding or removing protons from an atom does what to the atom? _____.
4. An atom with the same number of protons and electrons has a charge of _____.
5. Adding two electrons to a neutral atom produces an ion with a charge of _____.
6. An atom with six protons and five electrons would have a charge of _____.
7. What atom is created with nine protons, nine neutrons, and nine electrons?
8. Show the full symbol for the above atom in the box at the right
9. What does the upper-left number in the symbol represent? _____
10. What does the lower-left number in the symbol represent? _____



Game The Game

Click the tab at the top that says "Game"

Play each level until you make a perfect score. Call me over when you complete a level with a perfect score, and I will

check it off.

Level 1 _____

Level 2 _____

Level 3 _____

Level 4 _____

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N Noddings



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Overcoming Students' Misconceptions in Science Mageswary Karpudewan, Ahmad Nurulazam Md Zain, A.L. Chandrasegaran, 2017-02-28 This book discusses the importance of identifying and addressing misconceptions for the successful teaching and learning of science across all levels of science education from elementary school to high school It suggests teaching approaches based on research data to address students common misconceptions Detailed descriptions of how these instructional approaches can be incorporated into teaching and learning science are also included The science education literature extensively documents the findings of studies about students misconceptions or alternative conceptions about various science concepts Furthermore some of the studies involve systematic approaches to not only creating but also implementing instructional programs to reduce the incidence of these misconceptions among high school science students These studies however are largely unavailable to classroom practitioners partly because they are usually found in various science education journals that teachers have no time to refer to or are not readily available to them In response this book offers an essential and easily accessible guide

Sciences for the IB MYP 4&5: By Concept Paul Morris, Radia Chibani, El Kahina Meziane, Anna Michaelides, 2018-08-13 Develop your skills to become an inquiring learner ensure you navigate the MYP framework with confidence using a concept driven and assessment focused approach to Sciences presented in global contexts Develop conceptual understanding with key MYP concepts and related concepts at the heart of each chapter Learn by asking questions for a statement of inquiry in each chapter Prepare for every aspect of assessment using support and tasks designed by experienced educators Understand how to extend your learning through research projects and interdisciplinary opportunities Think internationally with chapters and concepts set in global contexts

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