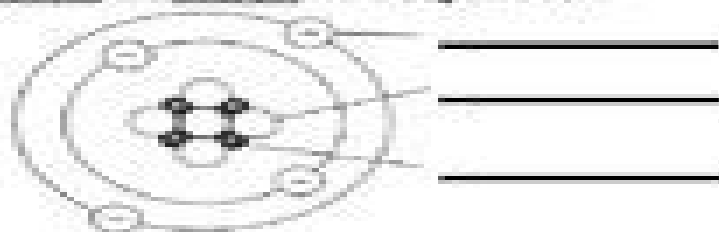


Chemistry of Life Review

1. Complete the table below:

Name	Symbol	Atomic Number	Number of Protons	Number of Electrons	Number of Neutrons
Carbon					
	O				
Hydrogen					
Nitrogen					

2. Label the protons, neutrons, and electrons in the diagram below.



3. Match each word with the definition.

- | | |
|---|--------------|
| _____ 1. Anything that occupies space and has mass. | A. acid |
| _____ 2. A substance made of two or more elements. | B. base |
| _____ 3. Pairs off an element with different numbers of neutrons. | C. carbon |
| _____ 4. The smallest amount of a substance. | D. catalyst |
| _____ 5. A homogeneous mixture of one or more solutes dissolved in a solvent. | E. compound |
| _____ 6. A substance into which another substance dissolves to form a solution. | F. inorganic |
| _____ 7. A substance being dissolved in a mixture or solution. | G. isotopes |
| _____ 8. A measure of acidity/basicity of a substance. | H. molecule |
| _____ 9. A substance with weak attraction ions. | I. organic |
| _____ 10. A substance with four H+ ions. | J. pH |
| _____ 11. A carbon-based substance in living systems. | K. solute |
| _____ 12. A substance that does not contain carbon. | L. solution |
| _____ 13. The basic element in organic compounds. | M. solvent |
| _____ 14. A substance that speeds up a chemical reaction. | N. cellulose |
| _____ 15. A polysaccharide in plants that serves as building material. | O. matter |

Answer Key To The Chemistry Of Life

S Marginson



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