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**R Faragher**



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web the first two equations have been balanced for you alpha particle  $\alpha$   $^4_2\text{He}$  beta particle  $\beta$  gamma ray  $\gamma$  proton  $p$  neutron  $n$  0 positron  $\beta^+$  0 1 30 1 65 1 2 63 2 15 0 30 0 65 30 9 29 2 2 12 6 15 32 30

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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by

writing a balanced nuclear equation  $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + x$  because the mass number is the same on both sides of the equation  $x$  must have a

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web identify the missing particle in the following nuclear reaction  $^{10}_0\text{n} + ^{235}_{92}\text{U} \rightarrow ^{140}_{54}\text{Xe} + ^{92}_{38}\text{Sr} + ^{1}_{0}\text{n}$

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