

Description

Cigarette lighter
Cigarette lighter fuse

Electrical system

Fuel air control valve (FAC)

Radio

Speaker 1/2

Speaker 1/3

Speaker 1/4

Speaker 1/5

Current
rank

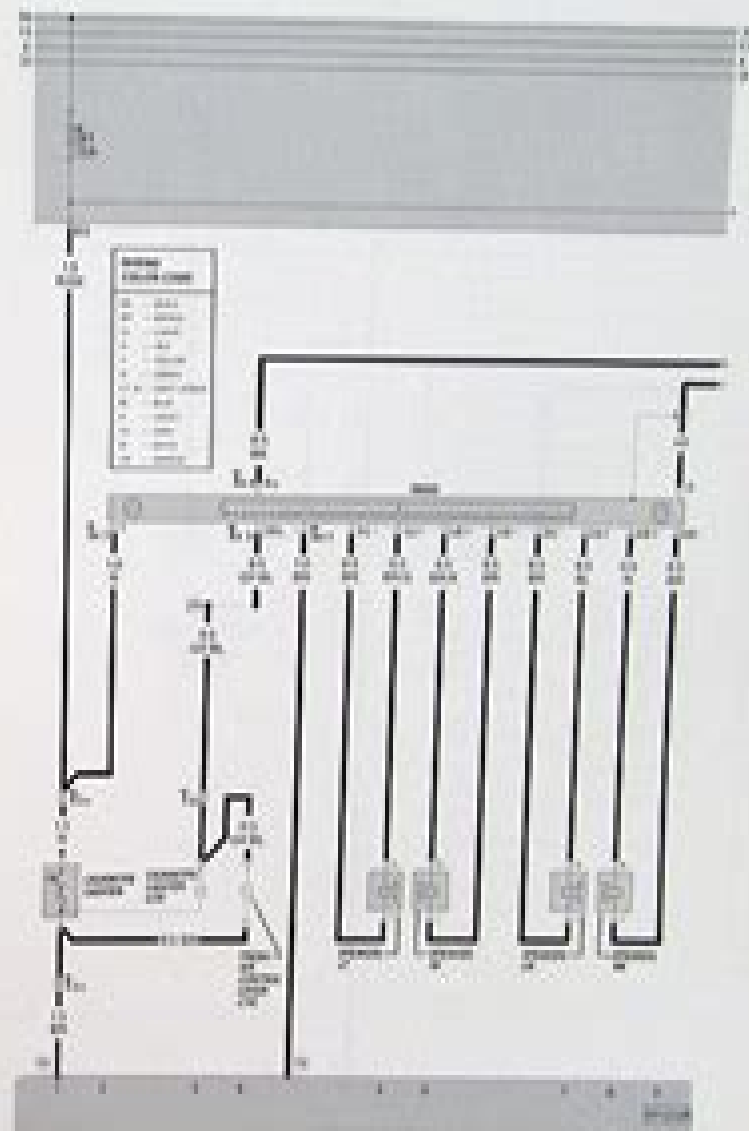
1	1
2	2
3	3
4	4
5	5
6	6
7	7
8	8
9	9
10	10

Wire connections

- 11a — single, under fuse relay panel
- 11b — single, under fuse relay panel
- 11c — double, under fuel panel, rear fuse relay panel
- 11d — double, under fuel panel, rear fuse relay panel
- 11e — double, under fuel panel, rear fuse relay panel
- 11f — double, under fuel panel, rear fuse relay panel
- 11g — double, under fuel panel, rear fuse relay panel
- 11h — double, under fuel panel, rear fuse relay panel
- 11i — double, under fuel panel, rear fuse relay panel
- 11j — double, under fuel panel, rear fuse relay panel
- 11k — double, under fuel panel, rear fuse relay panel
- 11l — double, under fuel panel, rear fuse relay panel
- 11m — double, under fuel panel, rear fuse relay panel
- 11n — double, under fuel panel, rear fuse relay panel
- 11o — double, under fuel panel, rear fuse relay panel
- 11p — double, under fuel panel, rear fuse relay panel
- 11q — double, under fuel panel, rear fuse relay panel
- 11r — double, under fuel panel, rear fuse relay panel
- 11s — double, under fuel panel, rear fuse relay panel
- 11t — double, under fuel panel, rear fuse relay panel
- 11u — double, under fuel panel, rear fuse relay panel
- 11v — double, under fuel panel, rear fuse relay panel
- 11w — double, under fuel panel, rear fuse relay panel
- 11x — double, under fuel panel, rear fuse relay panel
- 11y — double, under fuel panel, rear fuse relay panel
- 11z — double, under fuel panel, rear fuse relay panel

Ground connections

- 12a — rear fuse relay panel
- 12b — rear fuse relay panel
- 12c — rear fuse relay panel
- 12d — rear fuse relay panel
- 12e — rear fuse relay panel
- 12f — rear fuse relay panel
- 12g — rear fuse relay panel
- 12h — rear fuse relay panel
- 12i — rear fuse relay panel
- 12j — rear fuse relay panel
- 12k — rear fuse relay panel
- 12l — rear fuse relay panel
- 12m — rear fuse relay panel
- 12n — rear fuse relay panel
- 12o — rear fuse relay panel
- 12p — rear fuse relay panel
- 12q — rear fuse relay panel
- 12r — rear fuse relay panel
- 12s — rear fuse relay panel
- 12t — rear fuse relay panel
- 12u — rear fuse relay panel
- 12v — rear fuse relay panel
- 12w — rear fuse relay panel
- 12x — rear fuse relay panel
- 12y — rear fuse relay panel
- 12z — rear fuse relay panel



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d1 n d1 and we see we have the same sums so more generally if s_n d1 d2 d2 d1 n then the map s_n s_n defined by d1 d2 nd2 d1 n d1 is

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web oxidation reduction redox reactions oxidation $\text{Cu} + 2\text{H}^+ \rightarrow \text{Cu}^{2+} + \text{H}_2$ and reduction $\text{MnO}_4^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$ when the half reactions listed above are combined into a balanced chemical equation the ratio of the coefficients $\text{Cu} + \text{MnO}_4^- + 4\text{H}^+ \rightarrow \text{Cu}^{2+} + \text{MnO}_2 + 2\text{H}_2\text{O}$ is

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web we can balance it for charge by adding two electrons to the right side of the equation so that the net charge on each side is 0 oxidation $\text{Ni} \rightarrow \text{Ni}^{2+} + 2\text{e}^-$ now that the oxidation half reaction is balanced it tells us that two electrons are produced for every atom of nickel oxidized but where do those electrons go

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web balance each half reaction in basic solution 4 5 6 7 $\text{Cr}_2\text{O}_7^{2-} + 2\text{NO}_2 + 2\text{MnO}_2 + \text{Cr}_2\text{O}_3 + \text{NO}_3 + \text{SO}_2 + \text{Mn}_2\text{O}_3$ balance each redox reaction in acid solution using the half reaction method 8 $\text{H}_2\text{O}_2 + \text{Cr}_2\text{O}_7^{2-} + 9\text{TeO}_3 + 2\text{N}_2\text{O}_4 + \text{O}_2 + \text{Cr}_2\text{Te} + 10\text{ReO}_4 + 11\text{PbO}_2 + \text{IO}_2 + \text{Pb}_2 + 12\text{AsNO}_3 + \text{IO}_3 + \text{ReIO}_3 + \text{H}_2\text{AsO}_4$

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in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively $\text{C}_6\text{H}_8\text{O}_6 + 2\text{HNO}_2 \rightarrow \text{C}_6\text{H}_6\text{O}_6 + 2\text{H}_2\text{O} + 2\text{NO}$ choose 1 answer $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 a $\text{C}_6\text{H}_8\text{O}_6$ and NO_2 no 2 and $\text{C}_6\text{H}_8\text{O}_6$ b NO_2 and $\text{C}_6\text{H}_8\text{O}_6$ h and NO_2 c h and