

→ THE ATOMIC STRUCTURE

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1. From the following particles which one is neutral?
a. Proton
b. Electron
c. ✓ Neutron
d. None of these
2. From the following particles which one is positively charged?
a. Neutron
b. ✓ Proton
c. Electron
d. None of these
3. From the following particles which one is negatively charged?
a. ✓ Electron
b. Proton
c. Neutron
d. None of these
4. The smallest particle of matter is known as:
a. Electron
b. ✓ Atom
c. Molecule
d. Proton
5. Nucleus of an atom contains
a. Electron and Proton
b. ✓ Proton and Neutron
c. Electron and Neutron
d. All of these
6. The cathode rays experiment shows the presence of:
a. ✓ Electron in an atom
b. Neutron in an atom
c. Protons in an atom
d. Nucleus in an atom
7. In an atom Anode rays show the presence of:
a. Neutrons
b. ✓ Protons
c. Electrons
d. All of these
8. The elm ratio of an electron is:
a. 1.75×10^{09} coulomb/kg
b. 1.75×10^{08} coulomb/kg
c. 1.75×10^{10} coulomb/kg
d. ✓ 1.75×10^{11} coulomb/kg
9. The charge of an electron is:
a. ✓ 1.6022×10^{-19} coulomb
b. 1.6022×10^{-18} coulomb
c. 1.6022×10^{-16} coulomb
d. 1.6022×10^{-17} coulomb

Atomic Structure Mcq With Answers

Arshad Iqbal



Atomic Structure Mcq With Answers:

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