

Name: \_\_\_\_\_

Date/Period: \_\_\_\_\_

### ***Atomic Number and Mass Number***

Complete the following chart and answer the questions below.

<b><i>Element Name</i></b>	<b><i>Atomic Number</i></b>	<b><i>Number of Protons</i></b>	<b><i>Number of Neutrons</i></b>	<b><i>Mass Number</i></b>
carbon				12
	8		8	
hydrogen				1
		2		4
hydrogen			2	
nitrogen				14
			1	2
	92		146	
Potassium			20	
	11		12	
		47		108
Xenon			77	
			45	80
		24		52
			16	32
Gold				197
	76		114	

How are the *atomic number* and the *number of protons* related to each other?

How do the *number of protons*, *number of neutrons*, and the *mass number* relate to each other?

What is the *one thing* that determines the identity of an atom?

What sub atomic parts are equal to make a neutral atom?

Explain in your words how to calculate the mass of a sodium atom.

Draw and label three element symbols that are representative of the periodic table. It should have atomic #, mass #, name and element symbol.

# Atomic Number Activity Answer

**John Keighley, Stephen Doyle**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

## Atomic Number Activity Answer:

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