

Name: \_\_\_\_\_  
Date/Period: \_\_\_\_\_

### Atomic Number and Mass Number

Complete the following chart and answer the questions below.

| Element Name | Atomic Number | Number of Protons | Number of Neutrons | Mass Number |
|--------------|---------------|-------------------|--------------------|-------------|
| carbon       |               |                   |                    | 12          |
|              | 6             |                   | 6                  |             |
| hydrogen     |               |                   |                    | 1           |
|              |               | 1                 |                    | 4           |
| hydrogen     |               |                   | 2                  |             |
| nitrogen     |               |                   |                    | 14          |
|              |               |                   | 1                  | 2           |
|              | 92            |                   | 146                |             |
| Potassium    |               |                   | 20                 |             |
|              | 11            |                   | 12                 |             |
|              |               | 47                |                    | 108         |
| Xenon        |               |                   | 77                 |             |
|              |               |                   | 45                 | 80          |
|              |               | 24                |                    | 52          |
|              |               |                   | 16                 | 32          |
| Gold         |               |                   |                    | 197         |
|              | 76            |                   | 114                |             |

How are the atomic number and the number of protons related to each other?

How do the number of protons, number of neutrons, and the mass number relate to each other?

What is the one thing that determines the identity of an atom?

What sub atomic parts are equal to make a neutral atom?

Explain in your words how to calculate the mass of a sodium atom.

Draw and label three element symbols that are representative of the periodic table. It should have atomic #, mass #, name and element symbol.

## Atomic Number Activity Answer

**John Keighley, Stephen Doyle**

## Atomic Number Activity Answer:

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## **Table of Contents Atomic Number Activity Answer**

1. Understanding the eBook Atomic Number Activity Answer
  - The Rise of Digital Reading Atomic Number Activity Answer
  - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Number Activity Answer
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atomic Number Activity Answer
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Number Activity Answer
  - Personalized Recommendations

- Atomic Number Activity Answer User Reviews and Ratings
- Atomic Number Activity Answer and Bestseller Lists

5. Accessing Atomic Number Activity Answer Free and Paid eBooks

- Atomic Number Activity Answer Public Domain eBooks
- Atomic Number Activity Answer eBook Subscription Services
- Atomic Number Activity Answer Budget-Friendly Options

6. Navigating Atomic Number Activity Answer eBook Formats

- ePUB, PDF, MOBI, and More
- Atomic Number Activity Answer Compatibility with Devices
- Atomic Number Activity Answer Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Atomic Number Activity Answer
- Highlighting and Note-Taking Atomic Number Activity Answer
- Interactive Elements Atomic Number Activity Answer

8. Staying Engaged with Atomic Number Activity Answer

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Atomic Number Activity Answer

9. Balancing eBooks and Physical Books Atomic Number Activity Answer

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Atomic Number Activity Answer

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Atomic Number Activity Answer

- Setting Reading Goals Atomic Number Activity Answer
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Atomic Number Activity Answer

- Fact-Checking eBook Content of Atomic Number Activity Answer

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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