

Name: \_\_\_\_\_

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## S'mores Stoichiometry Lab

**Introduction:**

In this experiment, you will explore the principles of stoichiometry by building S'mores, the delicious, chocolate, marshmallow, and graham cracker treats.

**Definitions:** "Stoichio" means element and "metry" means the process of measuring. The mass and quantity relationships among reactants and products in a reaction are found using the process of stoichiometry.

**Purpose:**

To determine the limiting reactant in the synthesis of S'mores.

**Prelab Questions:**

1. What does a chemical equation tell us?
2. What do the coefficients in a balanced chemical equation represent?

**Materials:**

Miniature Chocolate bars

Graham cracker Square

Marshmallows

Paper Towels

Bunsen Burner

Wooden skewers

**Reference Data:**

Substance	Symbol	Unit mass	Package Mass
Graham Cracker Squares	Gr	8.50 g	403 g
Marshmallow	Mm	7.125 g	285 g
Miniature Chocolate Bar	Ch	12.00 g	343 g

\*\* for the purposes of this lab the UNIT mass and MOLAR mass are the same thing

**Procedure:**

1. Perform a synthesis reaction (forming one s'more). For this lab, we will consider a s'more to consist of 2 graham cracker squares, 1 marshmallow and 1 miniature chocolate bar.

- a) Determine the formula for a s'more
- b) write the balanced chemical equation using the symbols above
- c) According to the balanced equation complete the following ratios:

$$\begin{array}{rcl}
 \text{Gr} = & \text{Mm} & \text{Gr} = \text{S'more} \\
 \hline
 \text{Ch} = & \text{Mm} & \text{Mm} = \text{S'more} \\
 \hline
 \text{Ch} = & \text{Gr} & \text{Ch} = \text{S'more}
 \end{array}$$

2. Calculating the unit mass of the s'more

- a) Calculate the mass on ONE s'more.
- b) Is there a relationship between the mass of a S'more and the masses of the reactants used to make it?
- c) If so, what is the relationship?
- d) What law have you studied in this course that might define this relationship?

3. Predicting the limiting reactant:

- a) Complete the data table below based on the materials provided for your group.

Substance	Quantity received	# of S'mores it COULD make
Graham Cracker Squares		
Marshmallow		
Miniature Chocolate Bar		

- b) What will the limiting reactant be?
- c) How many of the other substances will remain?
- d) What are the masses of each excess reactant that will remain once the reaction is complete?

# Apply Stoichiometry Lab Answer Key

**M Woodhall**

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