



# Agrave Fleur De Peau

**Mark Steyvers**



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web complete the changes in concentrations for each of the following reactions  
 $\text{BaSO}_4 \text{ s} + \text{Ba}^{2+} \text{ aq} + \text{SO}_4^{2-} \text{ aq} \rightleftharpoons \text{Ba}^{2+} \text{ aq} + \text{SO}_4^{2-} \text{ aq}$   
 $\text{Al}(\text{OH})_3 \text{ s} + 3 \text{ aq} \rightleftharpoons \text{Al}^{3+} \text{ aq} + 3 \text{ OH}^{-} \text{ aq}$   
 $\text{Pb}(\text{OH})\text{Cl s} + \text{Pb}^{2+} \text{ aq} + \text{OH}^{-} \text{ aq} \rightleftharpoons \text{Pb}^{2+} \text{ aq} + \text{OH}^{-} \text{ aq}$

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