

Name \_\_\_\_\_ Group \_\_\_\_\_

Chemistry 2<sup>nd</sup> Semester  
Review 4th Period exam

1.- Identify as organic or inorganic

- a) KCl    b) C<sub>2</sub>H<sub>6</sub>O    c) CaCl<sub>2</sub>    d) CO<sub>2</sub>    e) NaCN    f) CH<sub>4</sub>

2.- Identify the following properties as organic or inorganic

- a) soluble in water    b) burns in air    c) low boiling point    d) contain K  
e) form ionic bond    f) form isomers    g) high boiling point    h) can be used as fuel

3.- True or false

- a) Carbon atoms can bond with other carbon atoms any number of times  
b) Carbon atomic number is 4  
c) Carbon forms 4 covalent bonds  
d) Carbon can bond with many elements, metals and nonmetals

4.- Identify the following as alkane, alkene or alkyne

- a)  $\begin{array}{c} \text{CH}_3\text{CH}_3 \\ | \\ \text{HC}\equiv\text{CCH}_2\text{CH}_2 \\ | \\ \text{CH}_3 \end{array}$     b)  $\begin{array}{c} \text{CH}_3-\text{CH}_2-\text{CH}-\text{CH}_2-\text{CH}_2-\text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$     c) CH<sub>2</sub>=CH-CH<sub>3</sub>

5.- Identify as saturated or unsaturated

- a)  $\begin{array}{c} \text{CH}_3-\text{CH}-\text{C}=\text{CH}_2 \\ | \quad | \\ \text{CH}_3 \quad \text{CH}_3 \end{array}$     b)  $\begin{array}{c} \text{H}_2\text{C}-\text{CH}-\text{CH}_3 \\ | \\ \text{CH}_3 \end{array}$

6.- Give the name of the following alkyl groups

- a) -CH<sub>3</sub>    b) -CH<sub>2</sub>-CH<sub>3</sub>    c) -CH<sub>2</sub>-CH<sub>2</sub>-CH<sub>3</sub>    d) -CH<sub>3</sub>-CH-CH<sub>3</sub>

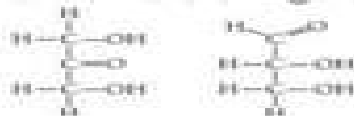
7.- What are aliphatic hydrocarbons.

8.- Identify as hydrocarbon or derivate of hydrocarbon

- a) CH<sub>3</sub>-Cl    b) CH<sub>3</sub>-CH<sub>2</sub>-CH<sub>3</sub>    c) CH<sub>3</sub>-COOH    d) CH<sub>3</sub>-CH=CH-CH<sub>3</sub>  
e) CH<sub>3</sub>-O-CH<sub>3</sub>    f) C<sub>6</sub>H<sub>6</sub>    g) CH<sub>3</sub>-CH=O

9.- What are isomers?

10.- Do the following molecules represent isomers? Explain.



11.- Circle the functional group and identify the class or family.

# Nd Semester Exam Chemistry Form

**Detroit (Mich.). Board of Education**



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### **Table of Contents Nd Semester Exam Chemistry Form**

1. Understanding the eBook Nd Semester Exam Chemistry Form
  - The Rise of Digital Reading Nd Semester Exam Chemistry Form
  - Advantages of eBooks Over Traditional Books
2. Identifying Nd Semester Exam Chemistry Form
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Nd Semester Exam Chemistry Form
  - User-Friendly Interface
4. Exploring eBook Recommendations from Nd Semester Exam Chemistry Form
  - Personalized Recommendations

- Nd Semester Exam Chemistry Form User Reviews and Ratings
- Nd Semester Exam Chemistry Form and Bestseller Lists
- 5. Accessing Nd Semester Exam Chemistry Form Free and Paid eBooks
  - Nd Semester Exam Chemistry Form Public Domain eBooks
  - Nd Semester Exam Chemistry Form eBook Subscription Services
  - Nd Semester Exam Chemistry Form Budget-Friendly Options
- 6. Navigating Nd Semester Exam Chemistry Form eBook Formats
  - ePub, PDF, MOBI, and More
  - Nd Semester Exam Chemistry Form Compatibility with Devices
  - Nd Semester Exam Chemistry Form Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Nd Semester Exam Chemistry Form
  - Highlighting and Note-Taking Nd Semester Exam Chemistry Form
  - Interactive Elements Nd Semester Exam Chemistry Form
- 8. Staying Engaged with Nd Semester Exam Chemistry Form
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Nd Semester Exam Chemistry Form
- 9. Balancing eBooks and Physical Books Nd Semester Exam Chemistry Form
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Nd Semester Exam Chemistry Form
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Nd Semester Exam Chemistry Form
  - Setting Reading Goals Nd Semester Exam Chemistry Form
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nd Semester Exam Chemistry Form
  - Fact-Checking eBook Content of Nd Semester Exam Chemistry Form

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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