

Name: _____

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Student Exploration: Limiting Reactants

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Keywords: chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a sausage and a bun. At the store, you buy four packages of eight sausages and three bags of 10 buns.

1. How many total hot dogs can you make? _____
2. Which ingredient limited the number of hot dogs you could make? _____
3. Which ingredient will you have leftover? _____

Gizmo: Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the **Limiting Reactants Gizmo™**, you can determine which reactant is limiting in various scenarios.



To begin, make sure $\text{H}_2 + \text{O}_2$ becomes H_2O is selected. The small "2" in H_2 , O_2 , and H_2O is a **subscript**. Subscripts represent the number of atoms in a molecule.

1. Use the sliders to set the number of O_2 molecules and H_2 molecules to two.
 - A. How many hydrogen molecules (H_2) are there? _____ Hydrogen atoms? _____
 - B. How many oxygen molecules (O_2) are there? _____ Oxygen atoms? _____
2. How many H_2O molecules do you think will form when these four molecules react? _____
3. Click Play (). How many H_2O molecules actually formed? _____

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Answer Key Gizmo

RD Boyd

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Table of Contents Full Version Student Exploration Limiting Reactants Answer Key Gizmo

1. Understanding the eBook Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - The Rise of Digital Reading Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Advantages of eBooks Over Traditional Books
2. Identifying Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - User-Friendly Interface

4. Exploring eBook Recommendations from Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Personalized Recommendations
 - Full Version Student Exploration Limiting Reactants Answer Key Gizmo User Reviews and Ratings
 - Full Version Student Exploration Limiting Reactants Answer Key Gizmo and Bestseller Lists
5. Accessing Full Version Student Exploration Limiting Reactants Answer Key Gizmo Free and Paid eBooks
 - Full Version Student Exploration Limiting Reactants Answer Key Gizmo Public Domain eBooks
 - Full Version Student Exploration Limiting Reactants Answer Key Gizmo eBook Subscription Services
 - Full Version Student Exploration Limiting Reactants Answer Key Gizmo Budget-Friendly Options
6. Navigating Full Version Student Exploration Limiting Reactants Answer Key Gizmo eBook Formats
 - ePUB, PDF, MOBI, and More
 - Full Version Student Exploration Limiting Reactants Answer Key Gizmo Compatibility with Devices
 - Full Version Student Exploration Limiting Reactants Answer Key Gizmo Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Highlighting and Note-Taking Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Interactive Elements Full Version Student Exploration Limiting Reactants Answer Key Gizmo
8. Staying Engaged with Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Full Version Student Exploration Limiting Reactants Answer Key Gizmo
9. Balancing eBooks and Physical Books Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Full Version Student Exploration Limiting Reactants Answer Key Gizmo
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Setting Reading Goals Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Fact-Checking eBook Content of Full Version Student Exploration Limiting Reactants Answer Key Gizmo
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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