

## Atomic Theory & Atom Structure: Multiple Choice

1. Which one of the following is not one of the postulates of Dalton's atomic theory?

- AD Atoms are composed of protons, neutrons, and electrons.
- BD All atoms of a given element are identical; the atoms of different elements are different and have different properties.
- CD Atoms of an element are not changed into different types of atoms by chemical reactions.
- ED Atoms are neither created nor destroyed in chemical reactions.
- ED Compounds are formed when atoms of more than one element combine; a given compound always has the same relative number and kind of atoms.
- ED Each element is composed of extremely small particles called atoms.

ANS: A

1. Consider the following selected postulates of Dalton's atomic theory.

- CD Each element is composed of extremely small particles called atoms.
- CD Atoms are indivisible.
- CD Atoms of a given element are identical.
- CD Atoms of different elements are different and have different properties.

2. Which of the postulates is/are no longer considered valid?

- AD (D) and (E)      BI (C) only      CI (B) and (D)      DI (B) only      EI (D) and (E)

ANS: C

1. All atoms of a given element have the same \_\_\_\_\_.

- AD mass      BI number of protons      CI number of neutrons
- ED number of electrons and neutrons      EI electrons

ANS: D

1. The nucleus of an atom does not contain \_\_\_\_\_.

- AD protons      BI protons or neutrons      CI neutrons
- ED subatomic particles      EI electrons

ANS: E

1. The nucleus of an atom contains \_\_\_\_\_.

# Atomic Structure Multiple Choice With Answers

**L Cohen**

## Atomic Structure Multiple Choice With Answers:

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