

Atomic Theory & Atom Structure: Multiple Choice

1. Which one of the following is not one of the postulates of Dalton's atomic theory?

- AD Atoms are composed of protons, neutrons, and electrons.
- BD All atoms of a given element are identical; the atoms of different elements are different and have different properties.
- CD Atoms of an element are not changed into different types of atoms by chemical reactions.
- ED Atoms are neither created nor destroyed in chemical reactions.
- ED Compounds are formed when atoms of more than one element combine; a given compound always has the same relative number and kind of atoms.
- ED Each element is composed of extremely small particles called atoms.

ANS: A

2. Consider the following selected postulates of Dalton's atomic theory.

- CD Each element is composed of extremely small particles called atoms.
- CD Atoms are indivisible.
- CD Atoms of a given element are identical.
- CD Atoms of different elements are different and have different properties.

3. Which of the postulates is/are no longer considered valid?

- AD C and D
- BD C only
- CD C and D
- ED C and D only
- ED C and D

ANS: C

4. All atoms of a given element have the same _____.

- AD mass
- BD number of protons
- CD number of neutrons
- ED number of electrons
- ED density

ANS: D

5. The nucleus of an atom does not contain _____.

- AD protons
- BD protons or neutrons
- CD subatomic particles
- ED electrons

ANS: B

6. The nucleus of an atom contains _____.

Atomic Structure Multiple Choice Answers

Christian Drosten

Atomic Structure Multiple Choice Answers:

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