

# Atomic Structure



Atomic Number

Element Symbol

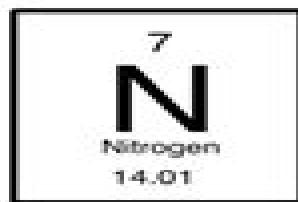
Element Name

Atomic Mass

The atomic number represents these two parts of the atom:  
number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons:

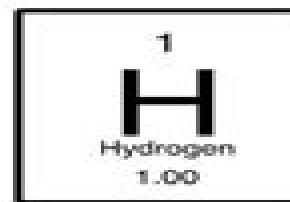
7

Electrons:

7

Neutrons:

7



Protons:

1

Electrons:

1

Neutrons:

0



Protons:

8

Electrons:

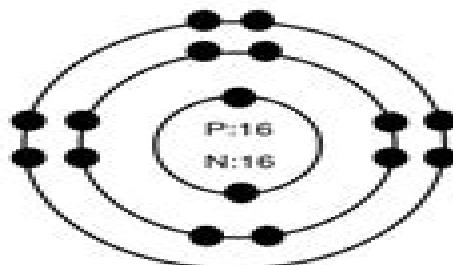
8

Neutrons:

8

## BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
  1. The first energy level can hold 2 electrons
  2. The second energy level can hold 8 electrons
  3. The third energy level can hold 8 electrons

Element: SulfurNumber of Valence Electrons: 6

The electrons in the outermost energy level are called:

Valence Electrons

## LEWIS DIAGRAMS

A Lewis diagram shows only the symbol with the Valence electrons



# Atomic Answer Key

**Lei Huang**

## Atomic Answer Key:

[1500 Science Test Questions/Answers](#) Dennis Arden Hooker,2025-01-01 1500 Science Test Questions w Keys Answers Statistical Analysis For Science Teachers Upper Elementary to College Dr Hooker researched and developed a book of 1500 Science Test Questions together with the Bloom's Taxonomy Discrimination Index the Key etc The book was funded through the National Science Foundation for teachers of Upper Middle School through College Science Programs 1500 Science Test Questions is an excellent tool for teachers to develop their own tests and for students to study for High School and College proficiency exams

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