





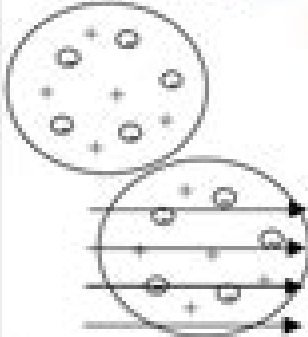
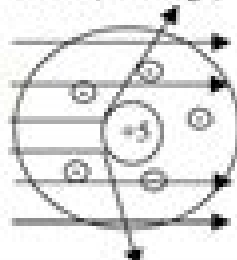


Atomic Theory Timeline

Democritus	John Dalton	Michael Faraday	J. J. Thomson	Robert Millikan	Ernest Rutherford
<p>~450 BC</p> 	<p>1803</p> 	<p>1839</p> 	<p>1896</p> 	<p>1909</p> 	<p>1909</p> 
Greek philosopher	English schoolteacher	English chemist	English physicist	American physicist University of Chicago	New Zealand scientist
<p>All matter around us is made of indivisible tiny particles- "atoms"</p> <p>(p.91)</p>	<p><u>Dalton's Atomic Theory</u></p> <ol style="list-style-type: none"> 1) Elements are made of tiny particles called <u>atoms</u>. 2) Atoms of one element are <u>identical</u> while atoms of different elements are <u>different</u>. 3) <u>Conservation</u> of atoms—rearrangement in RXN (Lavoisier previously stated this in terms of the Law of Conservation of Matter) 4) Different atoms form compounds in <u>constant ratios</u>. (Proust previously stated this in terms of the constant mass ratios) (p.92) 	<p>The structure of atoms is somehow related to <u>electricity</u>.</p> <p>(p.95)</p>	<p>Discovered atoms have negative particles (electrons) using a cathode ray tube.</p> <p>Discovered electron's charge to mass ratio: $1.76 \times 10^8 \text{ C/g}$</p> <p>(p. 97-98)</p> <p><u>Thomson's Plum Pudding Model, 1904</u></p> <p>Electrons are dispersed in a <u>uniform</u> positive charge. (p. 62 & 101)</p> 	<p>Measured the <u>charge</u> of an electron using oil droplets.</p> <p>Electron's charge: $1.60 \times 10^{-19} \text{ C}$</p> <p>Electron's mass: $9.11 \times 10^{-28} \text{ g}$</p> <p>(p. 98)</p> <p style="text-align: center;">Versus</p>	<p><u>Rutherford's Nucleus Theory</u></p> <p>Positive charge is not like a pudding, but concentrated in the <u>nucleus</u> as shown in the <u>Gold Foil (alpha particle) experiment</u></p> <p>*Most of an atom is empty space</p> <p>(p. 100-102)</p> <p>*1919- named positive charge the <u>proton</u> (+1) *1932- Rutherford and James Chadwick discover <u>neutron</u> in nucleus (no charge)</p> 

Atomic Theory Worksheet Answers

Stewart C. Bushong

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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[The Science Teacher](#) ,2001

Literacy Practices in Transition Anne Pitkänen-Huhta,Lars Holm,2012-11-14 Literacy Practices in Transition explores the connections between local situated literacy practices and global processes of mobility in the geographical space of the Nordic countries an example of contemporary mobile societies The detailed empirical analyses show how these connections affect individuals practices and policies how the global and local meet in discourses and practices and how people need to re negotiate their way in the complex and messy spaces in which they move The volume challenges current trends in the global standardization of language and literacy education Instead it promotes the idea of literacy as a multiple multilingual multimodal and constantly contestable and negotiable phenomenon which calls for the development of language and literacy education that is sensitive to the needs and experiences of the individual actors

Secondary Teaching Methods Kenneth T. Henson,1981

Teaching Science Online Dietmar Kennepohl,2023-07-03
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