

## UNIT 1: ATOMIC STRUCTURE

Periodic Table Address: \_\_\_\_\_

D. What do atoms have?

1. \_\_\_\_\_

**Example:** An atom has an atomic number of 8. How many protons are present in this atom?

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

**Now, try:**

Atoms of element Hydrogen have: \_\_\_\_\_

Atoms of element Oxygen have: \_\_\_\_\_

Atoms of element Iron have: \_\_\_\_\_

2. \_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

\_\_\_\_\_

**Example:** An atom has 16 protons and 16 neutrons. What is this atom's atomic mass number?

If an atom has a mass number of 20 and an atomic number of 8, how many neutrons are present in this atom?

# Atomic Structure Guided Notes Answers

**John R. Gordon**

## Atomic Structure Guided Notes Answers:

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