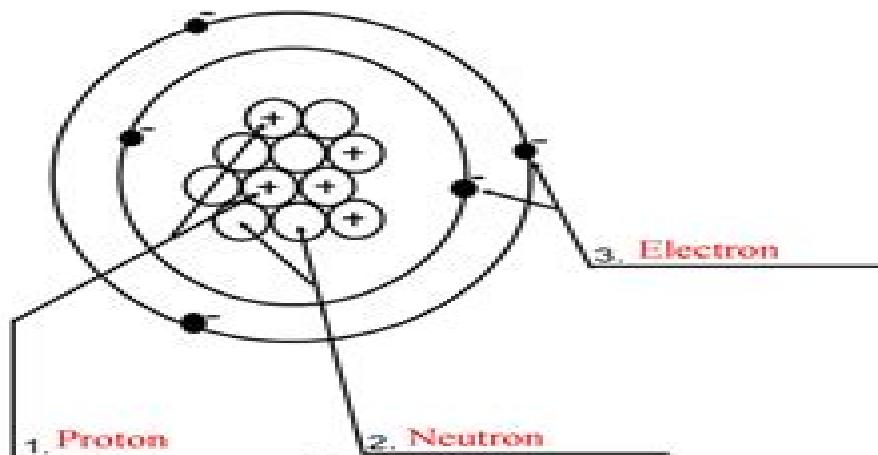


Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?
Positive (+1)
5. What type of charge does a neutron have?
Neutral (0)
6. What type of charge does an electron have?
Negative (-1)
7. Which two subatomic particles are located in the nucleus of an atom?
Protons and neutrons.

8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?
35
9. What is the atomic number of the atom in the diagram above?
5
10. What is the atomic mass/mass number of the atom in the diagram above?
(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number
11. How many protons are in the nucleus of an atom with an atomic number of 15?
15 *number of protons is the same as the atomic number
12. How many electrons are in the nucleus of an atom with an atomic number of 20?
20
13. How many neutrons are in the nucleus of an atom with an atomic number of 25? (use Periodic Table for mass)
30 Neutrons. atomic mass: 54.938 – 55 atomic number: 25....55-25=30
14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?
Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.
15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?
Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Atomic Structure Answers Chemistry

A Loxley

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Within the pages of "**Atomic Structure Answers Chemistry**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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 - The Rise of Digital Reading Atomic Structure Answers Chemistry
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Answers Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Answers Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Answers Chemistry
 - Personalized Recommendations
 - Atomic Structure Answers Chemistry User Reviews and Ratings

- Atomic Structure Answers Chemistry and Bestseller Lists
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 - Atomic Structure Answers Chemistry Budget-Friendly Options
- 6. Navigating Atomic Structure Answers Chemistry eBook Formats
 - ePUB, PDF, MOBI, and More
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 - Atomic Structure Answers Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Answers Chemistry
 - Highlighting and Note-Taking Atomic Structure Answers Chemistry
 - Interactive Elements Atomic Structure Answers Chemistry
- 8. Staying Engaged with Atomic Structure Answers Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Answers Chemistry
- 9. Balancing eBooks and Physical Books Atomic Structure Answers Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Answers Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Atomic Structure Answers Chemistry
 - Setting Reading Goals Atomic Structure Answers Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Atomic Structure Answers Chemistry
 - Fact-Checking eBook Content of Atomic Structure Answers Chemistry
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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