

Atomic Structure Worksheet

1 How many protons and neutrons are present in an atom of calcium (Mass number = 40)?

Protons = _____

Neutrons = _____

2 How many protons and neutrons are present in an atom of krypton (Mass number = 84)?

Protons = _____

Neutrons = _____

3 What is the name, symbol, and charge of an ion that contains 12 protons and 10 electrons?

Name = _____

Symbol = _____

Charge = _____

4 What is the name, symbol, and charge of an ion that contains 15 protons and 18 electrons?

Name = _____

Symbol = _____

Charge = _____

5 A neutral atom has 85 protons and 125 neutrons. Show calculations wherever necessary.

a What is its atomic number? _____

b What is its mass number? _____

c How many electrons does the atom have? _____

d What is the name of the atom? _____

6 Assuming all the atoms have neutral charge, complete the following table using your knowledge on atomic structure.

	Protons	Electrons	Neutrons	Mass number
Atom 1		8	8	
Atom 2			14	28
Atom 3	22			48
Atom 4		47		107
Atom 5		83	126	
Atom 6	87			223

Atom Structure Workbook Answers

Herbert Drapkin

Atom Structure Workbook Answers:

Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022 Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Sengupta Printed by IECIT Printing and publication Services Resource Centre Arabinda Nagar Bankura 722101 W B India Topics Covered Atoms and Molecules Structure of Atom Bonding Mechanism and Chemical Reactions Mechanism of Bonding This book is suitable for students of Class 9 to Class 11 Students aspiring for Pre Medical Entrance Examination can also get adequate support Additional Hard Copies can also be obtained from Chandan Sukumar Sengupta Arabinda Nagar Bankura 722101 WB Write to Us for more materials College Chemistry Questions and Answers PDF Arshad Iqbal, The College Chemistry Quiz Questions and Answers PDF Class 11 12 Chemistry Competitive Exam Questions Chapter 1 6 Practice Tests Grade 11 12 Chemistry Textbook Questions for Beginners includes revision guide for problem solving with hundreds of solved questions Class 11 12 Chemistry Questions and Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry Quiz PDF book helps to practice test questions from exam prep notes The Grade 11 12 Chemistry Quiz Questions and Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved tests Class 11 12 Chemistry Questions and Answers PDF Free download chapter 1 a book covers solved common questions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Chemistry Interview Questions and Answers PDF Download free eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The Class 11 12 Chemistry Interview Questions Chapter 1 6 PDF book includes college question papers to review practice tests for exams Class 11 12 Chemistry Practice Tests a textbook s revision guide with chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Questions Bank Chapter 1 6 PDF book covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure Questions Chapter 2 Basic Chemistry Questions Chapter 3 Chemical Bonding Questions Chapter 4 Experimental Techniques Questions Chapter 5 Gases Questions Chapter 6 Liquids and Solids Questions The Atomic Structure Quiz Questions PDF e Book Chapter 1 interview questions and answers on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model defects charge to mass ratio of electron discovery of electron discovery of neutron discovery of proton dual nature of matter electron charge electron distribution electron radius and energy derivation electron velocity electronic configuration of elements energy of revolving electron fundamental particles Heisenberg s uncertainty principle hydrogen spectrum magnetic quantum number mass of electron metallic crystals properties Moseley law neutron properties orbital concept photons wave number Planck s quantum theory properties of cathode rays properties of positive rays quantum numbers quantum theory Rutherford model of atom shapes of orbitals spin quantum number what is spectrum x rays and

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