

What happened to the systems of the Aral Sea in terms of inputs, outputs, and energy?

With respect to matter? What about energy?

Thermodynamics state? How does this relate to the Earth system?

Thermodynamics state? How does this relate to the Earth?

Thermodynamics, not all of the energy from the gasoline you put in forward. What happens to the rest?

Using (inputs vs. outputs), describe why our planet is heating.

What is the steady state of a system? What about positive

What is the steady state of a system? What about negative

What is the steady state of a system? What about negative

What is the steady state of a system? What about positive

What is the steady state of a system? What about negative

What is the steady state of a system? What about positive

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What is the steady state of a system? What about positive

What is the steady state of a system? What about negative

What is the steady state of a system? What about positive

What is the steady state of a system? What about negative



Water Pollution Review		
Pollutant (Example)	Source(s)	Problem(s)
SO ₂ : <ul style="list-style-type: none"> Oxygen requirement of bacteria decomposes dead organic matter includes agriculture 30%, SO₂ should be an insecticide 	<ul style="list-style-type: none"> Decomposition of dead organic matter Microbial degradation (SO₂) converted sulfur System 	<ul style="list-style-type: none"> Creates a dead zone Building salt and increases SO₂
Plastic (Garbage, oil, coal)	Plastic from mammals and birds, agriculture waste runoff (fertilizer) and human sewage	Decreases the level of oxygen depends on the type of trash collected

Environmental Science

Study Guide



Cheat Sheet

- What type of system is Earth, with respect to matter? What about energy?
- What is a closed system? (It is the atom that makes up the environment)
- What is an open system?
- What does the First Law of Thermodynamics state? How does this relate to energy? (Energy cannot be created or destroyed, it only changes from one form to another)
- What does the Second Law of Thermodynamics state? How does this relate to energy? (The entropy of the universe increases through nuclear fission and fusion, but it also uses energy. The energy is used to move oxygen.)
- What is the Law of Energy? Energy can neither be created or destroyed.
- What does the Second Law of Thermodynamics state? How does this relate to energy?
- What does the second law of thermodynamics state that you cannot break even. It changes from one form to another, it always moves from a more useful to a less useful form.
- Based on the second law of thermodynamics, not all of the energy from a car gets used to move the car forward. What happens to the rest? The rest can be good for air conditioning, and charging your phone.
- Using a systems analysis approach (inputs vs. outputs), describe what happened (global warming).
- Inputs (inputs) are absorbing and killing the carbons that maintain temperature. (output) is occurring.
- How does negative feedback affect the steady state of a system? What is feedback?
- A negative feedback is self-regulating and stabilizing because the output

Ap Environmental Science Jay Withgott Notes

G. TYLER. MILLER

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