

17

WATER AND AQUEOUS SYSTEMS

Conceptual Curriculum
 Concrete concepts
 More abstract concepts or
 mathematical problem-solving

Standard Curriculum
 Core content
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SECTION 17.1 LIQUID WATER AND ITS PROPERTIES (pages 475–478) 

This section explains the hydrogen bonding that occurs in water molecules and how high surface tension and low vapor pressure can be explained in terms of hydrogen bonding.

► The Water Molecule (pages 475–476)

- What unique molecular compound is the foundation of all life on Earth?
 water
- Approximately what fraction of Earth's surface is covered in water? 3/4
- Circle the letter next to each sentence that is true concerning water molecules.
 - Each O — H covalent bond in a water molecule is nonpolar.
 - In a water molecule, the less electronegative hydrogen atoms acquire a slight positive charge and the oxygen atom acquires a slight negative charge.
 - Because the water molecule has an H — O — H bond angle of 106°, the molecule as a whole is polar.
- The diagram below depicts a water molecule. Complete the labels showing the locations of the hydrogen atoms, the oxygen atom, and the regions of positive and negative charge.



- List five important properties of water that result from hydrogen bonding.

- high surface tension
- low vapor pressure
- high specific heat capacity
- high heat of vaporization
- high boiling point

1water And Aqueous Systems Guided Answers

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1water And Aqueous Systems Guided Answers:

Reviewing **1water And Aqueous Systems Guided Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing.

Within the pages of "**1water And Aqueous Systems Guided Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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