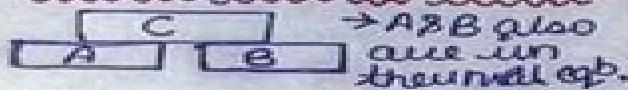


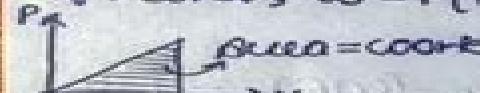
branch of science with conversion of heat and work.

* Zeroth Law of Thermodynamics



* First Law of Thermodynamics
 $\rightarrow dQ = dw + \Delta U$

• work done $\rightarrow dw = \int P dV$
 • if P const; $w = P(V_2 - V_1)$



$$\Delta U = \frac{f}{2} nR\Delta T$$

$$\{ \Delta U = nC_V\Delta T \}$$

$$\{ \Delta U = \frac{nR\Delta T}{Y-1} \} \frac{f}{2} = \frac{1}{Y-1}$$

$$\{ \Delta U = \frac{P_2V_2 - P_1V_1}{Y-1} \}$$

* Sign Conventions

→ Heat in \rightarrow +ve out \rightarrow -ve

$$\Delta U = \frac{f}{2} nR\Delta T$$

$$T \uparrow \rightarrow \Delta T \uparrow \therefore \Delta U \rightarrow +ve$$

$$T \downarrow \rightarrow \Delta T \downarrow \therefore \Delta U = -ve$$

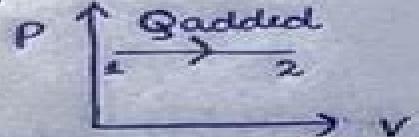
Q: $\{ T_{100} = 0 \}$ Find $\Delta U = ?$
 $\{ w = 60J \}$ $dQ = dw + \Delta U$
 $\{ \Delta U = 40J \}$

$$Q_{out} = -ve \quad Q_{in} = +ve$$

(i) Isobaric $\Rightarrow (P = C)$

$$PV = nRT \rightarrow V \propto T$$

$$\sqrt{T} = \text{constant}$$



Thermodynamics

$$w = \int P \cdot dV \Rightarrow w = P(V_2 - V_1)$$

$$\Delta U = nC_V\Delta T \quad \text{you can prove}$$

$$dq = nC_P\Delta T$$

(ii) Isochoric \Rightarrow $V = \text{constant}$

$$P \propto T \rightarrow P/T = \text{constant}$$

\rightarrow Gay-Lussac law.

$$P \uparrow \quad \{ w = 0 \}$$

$$\Delta U = nC_V\Delta T$$

$$dq = nC_V\Delta T$$

(iii) Isothermal (Boyle's law)

$$PV = \text{constant}, \quad XY = \text{constant}$$

$$P \uparrow \quad \{ w = P_1V_1 \ln \frac{V_2}{V_1} \}$$

$$P \uparrow \quad \{ w = nRT \ln \frac{V_2}{V_1} \}$$

$$\Delta U = 0 \quad dq = dw$$

(iv) Adiabatic $Q = 0$

\rightarrow Bursting of tyres

$$PV^Y = \text{constant}$$

$$Q = 0 \quad \{ \text{mono} = 1.66, \quad \text{dia} = 1.4, \quad \text{poly} = 1.33 \}$$

$$dw = -\Delta U$$

$$dw = nR(T_2 - T_1)$$

$$\Delta U = nC_V\Delta T \quad Y-1$$

$$\frac{P_2}{P_1} = \left(\frac{V_1}{V_2} \right)^Y, \quad \frac{T_2}{T_1} = \left(\frac{P_2}{P_1} \right)^{\frac{1}{Y-1}}$$

$$\frac{T_2}{T_1} = \left(\frac{V_1}{V_2} \right)^{Y-1}$$

$$(slope)_{baro} = Y \cdot (slope)_{isotherm}$$

$$dq = dw$$

$$(slope)_{adiab} = Y \cdot (slope)_{isotherm}$$

$$dq = dw$$

$$(slope)_{isobar} = Y \cdot (slope)_{isotherm}$$

$$dq = dw$$

$$(slope)_{isochor} = Y \cdot (slope)_{isotherm}$$

$$dq = dw$$

(v) Polytropic \Rightarrow

$$PV^n = \text{constant}$$

$$Q \neq 0$$

$$dq = dw + \Delta U$$

$$dw = nR(T_2 - T_1)$$

$$n-1$$

$$\Delta U = nRAT$$

$$Y-1$$

$$dq = \Delta U = nC_V\Delta T$$

$$P = \text{constant}$$

$$PV = \text{constant}$$

$$PV^Y = \text{constant}$$

$$V = \text{constant}$$

$$P = C$$

$$n=0$$

$$iso \rightarrow n=1$$

$$adi \rightarrow n=\infty$$

$$vol \rightarrow \infty$$

$$P \uparrow$$

$$T \uparrow$$

$$V \uparrow$$

$$P \uparrow$$

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1. Understanding the eBook 1std Physics Chemistry Short Notes Downld
 - The Rise of Digital Reading 1std Physics Chemistry Short Notes Downld
 - Advantages of eBooks Over Traditional Books
2. Identifying 1std Physics Chemistry Short Notes Downld
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 1std Physics Chemistry Short Notes Downld
 - User-Friendly Interface
4. Exploring eBook Recommendations from 1std Physics Chemistry Short Notes Downld
 - Personalized Recommendations
 - 1std Physics Chemistry Short Notes Downld User Reviews and Ratings
 - 1std Physics Chemistry Short Notes Downld and Bestseller Lists
5. Accessing 1std Physics Chemistry Short Notes Downld Free and Paid eBooks
 - 1std Physics Chemistry Short Notes Downld Public Domain eBooks
 - 1std Physics Chemistry Short Notes Downld eBook Subscription Services
 - 1std Physics Chemistry Short Notes Downld Budget-Friendly Options
6. Navigating 1std Physics Chemistry Short Notes Downld eBook Formats

- ePub, PDF, MOBI, and More
- 1std Physics Chemistry Short Notes Downld Compatibility with Devices
- 1std Physics Chemistry Short Notes Downld Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of 1std Physics Chemistry Short Notes Downld
- Highlighting and Note-Taking 1std Physics Chemistry Short Notes Downld
- Interactive Elements 1std Physics Chemistry Short Notes Downld

8. Staying Engaged with 1std Physics Chemistry Short Notes Downld

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers 1std Physics Chemistry Short Notes Downld

9. Balancing eBooks and Physical Books 1std Physics Chemistry Short Notes Downld

- Benefits of a Digital Library
- Creating a Diverse Reading Collection 1std Physics Chemistry Short Notes Downld

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine 1std Physics Chemistry Short Notes Downld

- Setting Reading Goals 1std Physics Chemistry Short Notes Downld
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of 1std Physics Chemistry Short Notes Downld

- Fact-Checking eBook Content of 1std Physics Chemistry Short Notes Downld
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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