

Name: _____

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Reading the Periodic Table **KEY**

1. List the atomic numbers of the elements in Period 2.

3, 4, 5, 6, 7, 8, 9, 10

2. List the symbols of the elements in Period 3.

Na, Mg, Al, Si, P, S, Cl, Ar

3. Name the elements in Period 2 that are metals.

Li, Be

4. Name the elements in Period 1 that are nonmetals.

H, He

5. Which element in Period 3 has the least mass?

Sodium (Na)

6. Which element in Period 4 has the most mass?

Krypton (Kr)

7. To which group/family do these belong?

a. Cobalt **transition metals**

b. Sodium **alkali metals**

c. Argon **noble gases**

d. Silicon **group 14 metalloids**

e. Chlorine **halogens**

f. Magnesium **alkaline earth metals**

8. Find iron on the periodic table:

a. What is the atomic number of iron? **26**

b. To which group does iron belong? **Transition metals**

9. Name the elements that have properties similar to that of chlorine.

Fluorine, bromine, iodine, astatine, tennessine

10. Fill in the chart below. The first row has been completed as an example.

Element	Symbol	Atomic Number	Period	Group
sodium	Na	11	3	1
oxygen	O	8	2	16
krypton	Kr	36	4	18
gold	Au	79	6	11
cadmium	Cd	48	5	12
manganese	Mn	25	4	7

Answers For Exploring The Periodic Table

Lei Shi



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Table of Contents Answers For Exploring The Periodic Table

1. Understanding the eBook Answers For Exploring The Periodic Table
 - The Rise of Digital Reading Answers For Exploring The Periodic Table
 - Advantages of eBooks Over Traditional Books
2. Identifying Answers For Exploring The Periodic Table
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Exploring The Periodic Table
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answers For Exploring The Periodic Table
 - Personalized Recommendations

- Answers For Exploring The Periodic Table User Reviews and Ratings
- Answers For Exploring The Periodic Table and Bestseller Lists
- 5. Accessing Answers For Exploring The Periodic Table Free and Paid eBooks
 - Answers For Exploring The Periodic Table Public Domain eBooks
 - Answers For Exploring The Periodic Table eBook Subscription Services
 - Answers For Exploring The Periodic Table Budget-Friendly Options
- 6. Navigating Answers For Exploring The Periodic Table eBook Formats
 - ePub, PDF, MOBI, and More
 - Answers For Exploring The Periodic Table Compatibility with Devices
 - Answers For Exploring The Periodic Table Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Exploring The Periodic Table
 - Highlighting and Note-Taking Answers For Exploring The Periodic Table
 - Interactive Elements Answers For Exploring The Periodic Table
- 8. Staying Engaged with Answers For Exploring The Periodic Table
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Exploring The Periodic Table
- 9. Balancing eBooks and Physical Books Answers For Exploring The Periodic Table
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Exploring The Periodic Table
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Exploring The Periodic Table
 - Setting Reading Goals Answers For Exploring The Periodic Table
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Exploring The Periodic Table
 - Fact-Checking eBook Content of Answers For Exploring The Periodic Table

- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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